Synthesis of Benzodioxepinone Analogues *via* a Novel Synthetic Route with Qualitative Olfactory Evaluation

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Marine odorants represent a minor yet diverse class of substances within the fragrance industry, of which 7-methyl-2H-1,5-benzodioxepin-3(4H)-one (1) is commercially known as *Calone 1951* $^{\circ}$, a synthetic first in the area of marine-fragrance chemistry. To determine the extent to which the characteristic marine odor of *Calone 1951* $^{\circ}$ corresponds to the substitution at the benzo portion of the molecule, a variety of aromatic substituents were incorporated into the benzodioxepinone structure (*Scheme 1, Table 3*). In light of the difficulty experienced in applying patented literature to deriving the analogues 12-18, particularly those with electron-withdrawing substituents, an alternative synthetic scheme was implemented for the construction of all analogues in favorable yields (*Scheme 4, Table 3*). Formation of the hydroxy-protected dihalo alkylating agent 24 *via* epoxide cleavage of epichlorohydrin (*Scheme 3*) allowed etherification favoring dihalo displacement and subsequent intramolecular ring closure ($\rightarrow 26a-g$). THP Deprotection followed by oxidation of the alcohols 27a-g to the ketones 12-18 provided a general pathway to the benzodioxepinone products. The influence of the substituent nature on odor activity revealed a diverse scope of olfactory character (*Table 4*).

Introduction. – The distinct marine-odor character and substantivity of *Calone* 1951® (=7-methyl-2*H*-1,5-benzodioxepin-3(4*H*)-one; 1), first synthesized by *Beereboom*, *Cameron*, and *Stephens* (*Pfizer*) [1] has contributed to its incorporation in the fragrance industry. *Calone* 1951® is used for seabreeze accords in fragrances such as 'Escape' (*C. Klein*, 1991), 'Polo Sport Woman' (*R. Lauren*, 1996), and 'Cool Water Woman' (*Davidoff*, 1997) [2]. Amongst traditional fragrance groups such as woody and floral, *Calone* 1951® has become an prominent compound in an atypical, formerly unestablished marine-fragrance category.

The novel properties of *Calone 1951*® render it suitable as a template for structure—odor-relationship (SOR) applications. The significant olfactory influence of the heterocycle O-atoms is evident in the contrast of odor character between the benzoxepinone 2 and the annulenone structures 3-5 or benzodioxepines 1 and 6. Some olfactory data for analogues of 1 are presented in *Table 1*.

Olfactory analysis of the benzoxepinone 2, also prepared by *Beereboom et al.* [1], was restricted to the term 'watermelon ketone', and α -keto-substituted derivatives of both 1 and 2 were assigned the same odor description. The annulenone analogue 3 was prepared by *Kubota* and *Isemura* in 1931 and described as exhibiting a bitter almond/peppermint odor [3]. More recently, benzocycloheptanones 4 and 5 were synthesized by *Yoshii* and co-workers [4] who defined them as possessing a lily-of-the-valley fragrance, with the presence of a marine accord in 5, the carbocyclic equivalent of 1.

Table 1. Molecular Features of some Benzodioxepinone Odorants

	X	Y	R	R^2 , R^3	Odor character
1	О	О	7-Me		marine, floral nuance
2	O	CH_2	Н		watermelon, green
3	CH_2	CH_2	Н		bitter almond, peppermint
4	CH_2	CH_2	7-⁴Bu		lily-of-the-valley
5	CH_2	CH_2	7-Me		lily-of-the-valley, marine
6	O	O	7- <i>n</i> -Pr		aldehydic
7	О	О	7-straight-chain and branched alkyl		marine, floral, aldehydic, fruity, with ozone, animalic, and ink-like appearances
8	О	О		H, Me	marine, aldehydic, floral, with citrus, nutty, and lichen facets

The 7-propyl-2*H*-1,5-benzodioxepin-3(4*H*)-one (6) was prepared by *Gaudin* and *Blanc* [5], and a range of other 7-substituted varieties 7 have been designed by *Kraft* [6] for which olfactory accords are dominated by marine, floral, aldehydic, and fruity notes. Tricyclic indenodioxepinone structures 8, also synthesized by *Kraft*, provided novel and reasonably intense marine analogues with the occasional citrus inflection, and singular nutty, moss, and animalic/civet appearances.

In our current studies pertaining to the odor properties of benzodioxepinone systems, we considered the modification of substitution and functionality at the aromatic ring and supported by qualitative olfactory analysis for consideration in ongoing SOR (structure – odor-relationship) studies by our group [7]. Preparation of benzodioxepinones substituted at the benzo moiety required introduction of an alternate broadly applicable synthetic pathway due to the diverse nature of the substituents. Limitations of the current approach (see below, *Scheme 1*) are examined, and the application of a simple but novel methodology towards benzodioxepinone syntheses (see below, *Scheme 4*) will be discussed.

Results and Discussion. – The current synthetic methodology for the preparation of benzodioxepinone structures is presented in *Scheme 1* (*Procedure A*) [1]. Our previous studies suggested that the *Williamson* ether synthesis conducted under microwave irradiation can lead to improvements in yield and purity of dietherified intermediates [8]. This is advantageous in the preparation of fragrance chemicals, as structurally related impurities are often difficult to remove and can persist in the benzodioxepinone products, causing erroneous olfactory evaluation. A range of benzodioxepinone analogues, *i.e.*, 12–18 were prepared under *Williamson* reaction conditions from the corresponding *ortho*-dihydroxy-substituted aromatics 9a-g (*Table 2*) and a 2 molequiv. excess of methyl bromoacetate (*Scheme 1*). Differences in reactivity were observed in the formation of diester derivatives 10a-g, depending on the method of

Scheme 1. Procedure A

a) K₂CO₃, BrCH₂COOCH₃, DMF. b) KO'Bu, THF. c) 2м HCl, aq. EtOH.

a) For R1, see Tables 2 (9a-g and 10a-g) and 3 (12-18)

heating used, *i.e.*, microwave or conventional heating (*Table 2*). Yields for **10d** and **10e** [9] obtained from the catechols (=benzene-1,2-diols) **9d** and **9e** containing electron-withdrawing substituents were improved with microwave heating. The degree of influence of substitution at the benzo moiety on the remainder of the molecule was evident in the NMR data of intermediates 10a-g, 11a-g (see below), and benzodioxepinones 12-18 (see below)¹).

Table 2. Microwave vs. Conventional Methods for the Williamson Etherification. Highest yields are given.

R^1	9	Yield [%] of 10	Purity [%] of 10 ^a)
3-Me	a	92 ^b)	95
3-MeO	b	66 ^b)	98
H	c	88 ^b)	98
4-NO ₂	d	74°)	100
4-CHO	e	79°)	91
4- ^t Bu	f	75 ^b)	98
(C(4))-CH=CH-CH=CH- $(C(5))$	g	96 ^b)	97

^{a)} Purity determined by GC/MS prior to column chromatography. ^{b)} Yield of isolated product obtained by the conventional method $(90-100^{\circ}, 2 \text{ h})$. ^{c)} Yield of isolated product obtained by the microwave method (200 W, 4 min).

Heteronuclear HMBC data verified the connectivity of the CH₂ and benzene moieties of all analogues of 10. The ¹H- and ¹³C-NMR chemical shifts of the CH₂ groups of 11 and target analogues 12-

¹⁾ Arbitrary atom numbering; for systematic names, see Exper. Part.

18 (see below) were established by HMQC and HMBC analysis. The nonequivalence of the two CH₂ and the two MeO groups in the 1 H- and 13 C-NMR spectra of 10 established that the inductive electronic effect of the R¹ substituent influences the entire molecular structure. As expected, this is evident in *ortho*-substituted derivatives 10a (δ (H) 4.69 (CH₂(1")) and 4.60 (CH₂(2))) and 10b (δ (H) 4.65 (CH₂(1")) and 4.64 (CH₂(2))¹)). The inductive effect on the terminal MeO groups from 4'-substitution is prominent in the NMR data of 10d (δ (H) 3.82 (Me(3)O) and 3.81 (Me(3")O); δ (H) 4.83 (CH₂(2)) and 4.80 (CH₂(1"))), whereby the lone pair of the NO₂ group enhanced electron-donating resonance effects. The inductive strength of the aldehyde moiety in 10e (δ (H) 4.81 (CH₂(1")) and 4.78 (CH₂(2))) did not match the effect of the nitro group and seemed restricted to the CH₂ groups, as did the influence of the *t*-butyl group in 10f (δ (H) 4.72 (CH₂(1")) and 4.68 (CH₂(2))).

The subsequent *Dieckmann* condensation reaction of the symmetrical hetero-ester chains of 10 resulted in regioisomer mixtures 11. Preliminary experiments have been published previously [8]. Our trials to form the β -keto esters returned varying yields of cyclized products 11 with KO'Bu in THF. Cyclization attempts to produce 11 mediated by the presence of a *Lewis* acid (TiCl₄) [10] gave no product after 48 h at 80°. The symmetrical side chains prevented ring closure despite the use of 0.5 mol-equiv. of TiCl₄ to encourage formation of the mono- rather than di-titanium enolate. Rosnati and De Marchi likewise reported unsuccessful efforts to prepare the oxobenzodioxepinecarboxylate structure by Dieckmann cyclization [11]. We noticed that the formation of isomers 11 with KO'Bu/THF required 10 to be \geq 98% pure; the presence of minor impurities led to decomposition and poor product yields (30-50%). The separation of the regioisomers was unnecessary since subsequent decarbomethoxylation of both isomers led to the desired benzodioxepinone products. The Dieckmann condensation for the cyclization of substituted diesterified catechols was not as general as we had anticipated. Diesters 10a - c, f provided good yields of cyclized material 11a - c, f (82 -87%); however, the yields of diesters 11d, e, g were irreproducible and complex product mixtures were formed.

The formation of bicyclic benzodioxane impurities 6- and 7-R¹-benzo[1,4]dioxin-2-one **20** [8] (*Scheme* 2), first noticed during the synthesis of the *Calone* 1951® precursors (R¹=Me) were the major recurring by-products appearing in highly variable yields in the product mixtures of the *Dieckmann* condensation. Product mixtures containing both **11** and the major impurities **20** (R¹=Me) are a viscous resin and difficult to separate, which *Carter et al.* [12] also discovered in their quest to synthesize acetic acid derivatives of catechol, resorcinol, and quinol. Rudimentary

Scheme 2. Major Impurities of the Dieckmann condensation of 10: 6- and 7-R1-benzo[1,4]dioxin-2-one

isomer 1: R¹ at C(6) (or C(5)) isomer 2: R¹ at C(7) (or C(8))

a) For R¹, see Table 2

isolation of **20a** ($\mathbb{R}^1 = \mathbb{M}e$) revealed it as a red resin, devoid of olfactory character, that forms a brittle solid on prolonged exposure to the atmosphere. Corresponding analogues 20a - g (for a - g, see Table 2) exhibited similar physical properties of which **20b** [13], **20c** [14], **20d** [14], **20f** [13], and **20g** [15] have previously been prepared. Initially it was suspected that the main impurity appearing in the synthesis of all analogues 11 were substituted derivatives 20. Complex cyclization mixtures obtained for 11d, e required separation by HPLC, and inconsistencies arose when NMR data of the isolated by-products were compared with GC/MS results. HPLC Separation was also employed to purify the target products 15, 16, and 18, establishing by subsequent NMR analysis that these impurities were the ether-cleavage products 21 (Fig. 1) [16]. Conclusions that KO'Bu-mediated Dieckmann condensation caused the formation of a mono-alkylated catechol that thermally cyclized in the GC injector port to the sterically favored 6-membered heterocyclic ring were concordant with both NMR and GC/MS data. On-column transesterification of 21 to the ethyl ester derivative provided a compound which was resistant to cyclization under GC conditions, thus confirming the formation of a mono-alkylated species from 10d, e under Dieckmann cyclization conditions.

 $R^1 = NO_2$, CHO

Fig. 1. Monoalkylated impurities present in the product mixtures containing 11d and 11e

We established by means of rudimentary ChemDraw models (MM2 and PM3 optimized), that aromatic substitution of the benzodioxepinone system has a significant influence on the geometric conformation adopted by the heterocyclic ring. The dependence of perceptible 1D-NMR shift differences on the kind and position of the functional groups in intermediates 10a-g and 11a-g and in benzodioxepinones 12-18 supported this. Archer and Claret [17] published results on the twist angle (θ) along the bond between the aromatic ring and the O-atoms of the heteroalicyclic ring obtained from the extinction coefficients in the UV spectra indicative of $\pi \to \pi^*$ transitions. Their modelling investigations led to the conclusion that the heteroalicyclic ring adopts either a more stable, rigid chair form or a flexible skewed pseudo twist-boat conformation, and that the boat conformation is only adopted by a negligible percentage of the compound due to steric constraints.

Semi-empirical calculations (MOPAC; AM1 followed by PM3) and *ab initio* models with HOMO/LUMO predictions were used to interpret differences in reactivity. Presenting compound **10d** as a representative in *Fig.* 2, the difficulty for compounds **10d** and **10e**, and to an extent for **10c** [18] and **10g**, to cyclize became evident. LUMO Hybridization is towed away from the alicyclic portion and concentrated around the aromatic ring, unlike those with electron-donating alkyl substituents as demonstrated by the 4-methyl isomer **10** ($R^1 = 4$ -Me; *Fig.* 3). Based on

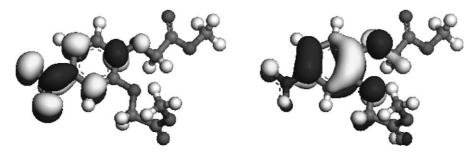


Fig. 2. LUMO (left) and HOMO (right) representations of the 4-nitro compound 10d (R¹ = 4-NO₂)

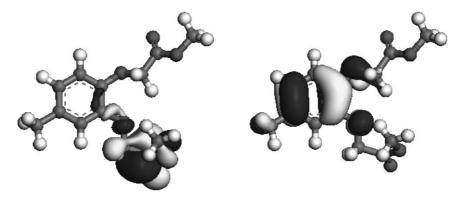


Fig. 3. LUMO (left) and HOMO (right) representations of compound the 4-methyl isomer $\mathbf{10}$ ($R^1 = 4$ -Me)

these models, intramolecular cyclization is substantially hindered for 10d due to a lack of available orbitals for acceptance of electrons required in the formation of a new σ bond. For the decarbomethoxylation (*Scheme 1*) in the absence of LUMO around the reaction site, the energy required for bond breaking and forming for the saponified side chain is much greater than that offered by external HOMO. Other factors such as polarization and potential tautomerism may also contribute. Electron-donating aromatic substituents present in compounds 10a, b, f assist in localizing the charge of the O-atoms and destabilization of the aromatic ring. Conversely the electronwithdrawing substituents of 10d, e and the conjugated naphthodioxepine system of 10g contribute to resonance stabilization of the aromatic ring, resulting in hindered cyclization. Of compounds 11a - g, the MeO derivative 11b is the only product obtained as a solid and the only intermediate for which the NMR data shows no distinguishable regioisomers, although GC/MS data reveals an 81:19 isomer ratio. The isomeric ratios for 11a, 11d, 11e, and 11f are 30:70, 57:43, 100 and 51:49, respectively. Signals due to the enol form of compounds 11a-g were not observed in the ¹H-NMR or FT-IR spectra, and CH and CH₂ signals were confirmed by DEPT-135, COSY, and HMQC analysis.

For the liberation of the target benzodioxepinones 12-18 (*Table 3*) a single-step acidic decarbomethoxylation of 11 (*Scheme 1*) afforded more desirable results than

Table 3. Yields of Target Analogues 12-18

$$R^1 = 0$$

	\mathbb{R}^1	Procedure A		Procedure B	
		Yield [%]a)	Purity [%]b)	Yield [%]a)	Purity [%]b)
12	6-Me	14	97	56	78
13	6-MeO	95	99	87	86
14	Н	10	87	60	98
15	$7-NO_2$	2	53	84	99
16	7-CHO	11	69	52	93
17	7-¹Bu	84	95	73	94
18	(C(7))-CH=CH-CH=CH- $(C(8))$	7	81	61	58

^a) Yield following purification with respect to **11** (*Procedure A*) or **27** (*Procedure B*). ^b) Purity by GC/MS prior to purification by column chromatography or HPLC.

initial saponification under basic conditions followed by acidic decarboxylation. In agreement with our electronic considerations, those variants **10** with an electron-withdrawing substituent at the aromatic moiety were more susceptible to nucleophilic alkoxy attack during the formation of **11d** and **11e**, and as a result, the latter were also less vulnerable to decarbomethoxylation. Purification of the decarbomethoxylated analogues **12–18** was dependent on the type of aromatic substituent present. Analogues with electron-donating aromatic substituents such as **12**, **13**, and **17** could be effectively purified by vacuum distillation, whereas those with an electron-withdrawing substituent or no electron-donating contributor, *i.e.*, **14** [19], **15**, **16**, and **18**, demanded separation by HPLC. Analysis of the ¹H-NMR data of the benzodioxepinone target analogues revealed a characteristic CH₂ s for the unsubstituted aromatic products **14** and **18**, and two separate CH₂ signals for compounds containing aromatic substitution (see *Exper. Part*).

The low yields of **15** and **16** (*Table 3*) and the inherent instability of compounds **11d** and **11e** required a more viable approach to the ketone ring to access the broader scope of analogues containing reactive or strongly electron-withdrawing aromatic substituents. Single-step trials to form the benzodioxepinone adduct by *Rosnati et al.* [11] have included the reaction between catechol and 1,3-dichloroacetone in the presence of *N,N*-dimethylpyridin-4-amine (DMAP) with no success. A study by these authors on the alkaline condensation of catechol with 1,3-dichloropropan-2-ol briefly described the formation of the 1,5-benzodioxepin ring as a minor impurity. Modification of this generally unutilized approach was applied to our current studies with the incorporation of a protected 1,3-dihalo alcohol. Our group has successfully applied this novel synthetic route (*Procedure B*) to generate analogues **12–18** (*Table 3*). Thus, 1-bromo-3-chloropropan-2-ol was prepared by acid-catalyzed epoxide opening of epichlorohydrin (**22**; *Scheme 3*) with HBr to give the secondary alcohol **23** exclusively. anti-*Markovnikov*-type regioselectivity is commonly encountered for these ambident substrates (*Shargi* and co-workers) [20]. The trimethylsilyl group was initially

Scheme 3

$$CI \longrightarrow OH \longrightarrow CI \longrightarrow Br \longrightarrow OOO$$

$$22 \qquad 23 \qquad 24$$

a) HBr (48%), r.t. b) DHP, P₂O₅, r.t.

implemented as a protecting group for **23**; however, it proved too labile for etherification reactions even at milder temperatures. Thus, the robust (under basic conditions) protecting group tetrahydropyran (THP) was introduced by reaction of **23** with 3,4-dihydro-2H-pyran (DHP) and P_2O_5 as catalyst under solvent-free conditions [21], to produce **24** as a distillable colorless oil in 91% yield.

The experimental approach employed in *Procedure B* (Scheme 4) involved initial reaction of the benzene-1,2-diols $9\mathbf{a} - \mathbf{g}$ with an equimolar amount of alkylating agent 24 to expediently form the stable monoalkylated intermediate 25 by chemoselective displacement of the Br-atom as shown by GC/MS. Intramolecular etherification of the second OH group with the chloroalkane moiety gave the cyclized 7-membered ring upon prolonged heating in DMF. Although the Cl-displacement required 24 h, no dimeric structures or polymeric material were formed. Initial trials involved the use of

Scheme 4. Procedure B

a) K_2CO_3 , DMF, reflux. b) V_2O_5 , H_2O_2 , 70° . c) KMnO₄, 4% KOH soln., r.t. (for $R^1 = NO_2$ or CHO, oxalyl chloride, DMSO, Et₃N, -78°).

a) For R^1 , see *Tables 2* (9a-g) and 3 (12-18)

NaH but produced inferior results to K₂CO₃ in DMF, which was sufficient to deliver clean mixtures of the cyclized products 26 in excellent yields. Deprotection of 26 with silica chloride [22] according to a protocol by Firouzabadi et al. was successful in the removal of THP, but was accompanied by significant decomposition. A trial of a dual oxidative deprotection with β -cyclodextrin (deprotection) and N-bromosuccinimide (NBS; oxidation) in H_2O /acetone [23] resulted in a small yield of the α -bromo carbonyl compound after 15 min, and with incorporation of a Br-scavenger returned only the starting alcohol. A KMnO₄/CuSO₄ hydrate with an alcohol catalyst [24] was attempted, also to no avail. A NaBrO₃/NH₄Cl reagent [25] in aqueous MeCN, recommended for oxidative deprotection of silyl ethers resulted in attack at the fusion atoms forming over-oxidized catechol derivatives. Deprotection of the THP group (Scheme 4) with V₂O₅ catalyst under conditions developed by *Patel* and co-workers [26] produced the alcohol in quantitative yield under mild conditions. The aldehyde derivative 26e was also successfully deprotected with V_2O_5 avoiding over-oxidation and polymerization. Esterification of the aldehyde was avoided by the use of MeCN rather than MeOH as solvent as suggested by Patel and co-workers. The ¹H-NMR signal for the proton at the stereogenic center C(3) of 26a – g characteristically overlapped with the diastereotopic CH₂(2) and CH₂(4) protons. The ambiguous ¹H-NMR resonance was clarified by COSY and HMQC experiments in each case.

Oxidation of the secondary alcohols **27** (*Scheme 4*) with KMnO₄ in benzene and under PTC-assisted biphasic conditions (PTC = phase-transfer catalyst) failed to provide the oxidized product in greater than 5% yield. Aqueous basic conditions suggested by *Rosnati et al.* were found to be optimal for the oxidative activity of KMnO₄ and provided the ketone products **12**–**14** and **17**–**18** at room temperature in good yields (*Table 3*). Surprisingly, the aqueous acidic workup resulted in chlorinated by-products when CH₂Cl₂ was used as extraction solvent, so Et₂O was used instead. Permanganate was not a suitable oxidizing agent for the formation of benzodiox-epinone analogues with electron-withdrawing aromatic substituents. Application of *Swern*-oxidation conditions by using oxalyl chloride and DMSO gave reproducible formation of the nitro and aldehyde derivatives **15** and **16** in 84% and 52%, respectively and in favorable purity (99% and 93%, resp., *Table 3*). HMQC and HMBC couplings established CH₂/fusion atom correlations for all target analogues (including **19**, see below) and the intermediates **26** and **27** [27].

Halogenated compounds rarely appear in the fragrance-chemistry arena, in part due to their inherent toxicity. In determining the importance of the aromatic moiety to the *Calone 1951*® headspace, the accessible aromatic halo-substitution product **19** provided an informative addition to the analogue group. Bromination of *Calone 1951*® (1) gave **19**, the single molecule in this series not derived from a catechol. This bromination was based on a system developed by *Kikuchi*, *Sakaguchi*, and *Ishii* [28] (*Scheme 5*), *i.e.*, a two-phase system for the *in situ* generation of hypobromous acid (HOBr) from NaBrO₃/NaHSO₃ in the aqueous phase, which is suspected to form a cationic bromine species. Bromination of the aromatic ring occurred selectively at the *ortho* position to the Me group as predicted by the activating Me group.

Olfactory Evaluation. – The diverse range of synthesized analogues incorporated many which maintained the original marine-odor attribute of 1, namely 12, 13, 17, and

19 and to a lesser extent 15 and 16 (*Tables 4* and 5). In all cases, the odor threshold was largely increased compared to 1, as indicated by a low perceived intensity for the range of analogues. Aromaticity, particularly as benzylic functionality is prevalent in compounds used for their fragrant properties. The patent released by *Pfizer* allocated a general olfactory assessment of 'watermelon green' to a range of benzodioxepinone

Table 4. Olfactory Data for Analogues 12-19

		Qualitative olfactory data ^a)
12	0 0	Fresh, floral-aldehydic odor of green and woody-marine tonality
13	OMe	Sweet-spicy, green-marine, floral, slightly earthy, walnut-like nuances, relatively weak
14	0 0	Green-floral, balsamic odor in the direction of Peru and Tolu balsam resinoid, a slight marine salty aspect
15	O_2N O O	Fruity-spicy, balsamic odor with a slight woody inflection and a distant powdery-vanillic sweetness, weak
16	OHC O O	Sweet, powdery scent reminiscent of vanilla with a spicy-balsamic undertone
17	¹ Bu 0 0	Fresh, marine, ozony odor, a green and walnut-like nuance, slightly floral
18	0 0	Oily, floral-green, and a slightly metallic odor, a very weak watery-floral, Calone-like inflection, weak
19	Br O O	Marine-ozony, ethereal, walnut-like odor, with salty aspects and a floral-fruity, geranium-like inflection, slightly earthy; same intensity and tenacity as 17

^a) Data provided by *Philip Kraft* and *Alain E. Alchenberger*, *Givaudan Switzerland AG*, Fragrance Research.

oily spicy

12 13 **17** 18 19 14 15 16 \mathbb{R}^1 Me CHO ^tBu 7-Me, 8-Br MeO Benzo marine green floral aldehyde woody/earthy nutty fruity vanillic balsamic metallic salty

Table 5. Graphic Representation of Olfactory Characteristics of Compounds 12-19. Black: present; grey: present but weak; white: absent.

a) NaBrO₃, NaHSO₃, H₂O/cyclohexane, 85%.

compounds including 14 and 17 of our work. Compound 14 and 18 exhibit similar odor profiles concordant with their comparable molecular attributes: a benzo and a naphtho fusion component, respectively. The green note is preserved, however, the marine aspect is weakened for both structures, indicating the importance of aromatic substitution to the typical Calone 1951® olfactory character. Interestingly, complete absence of substitution converts the fragrance emanated to a sweet array, particularly by the reference to balsam of Peru and Tolu that typically exhibit deep, rich sweetness. Both the nitro and aldehyde derivatives 15 and 16, respectively, displayed individual olfactory characteristics. The marine character is completely removed in both cases, with introduction of fruity, sweet, balsamic tonalities, with a hint of spiciness for 15. The steric bulk introduced by the 'Bu and Br groups in 17 and 19, respectively, may contribute to the walnut inflection perceived for both. Compound 18 serves as an interesting comparison, maintaining the marine aspect understandably as the entire Calone 1951® molecular skeleton is preserved for this analogue. The marine odor is steered in a more salty direction, accompanied by floral notes, as also often detected in the headspace from 1. To preserve the characteristic Calone 1951® odor, substitution is not restricted to the 7-position as demonstrated when comparing compounds 12, 13, 16, and 18. The type of functionality present at the aromatic ring, rather than its position, is more active in deconstructing the marine-odor character, as exhibited by the qualitative assessment of 14-16.

Conclusion. – The presence of the aromatic Me substituent in *Calone 1951*® complements but is not critical for the essential marine odor characteristic. In conjunction with the presence of the C=O group in the aliphatic ring which was previously studied [7], it is evident from our observations that aromatic-ring substitution modifies the odor perception of the marine character of *Calone 1951*®.

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Experimental Part

1. General. Potassium tert-butoxide and methyl bromoacetate were supplied by Merck. Otherwise, all reagents were obtained from Aldrich Chemical Co. and used without further purification. THF was dried over CaH₂, followed by distillation from sodium/benzophenone. DMF was dried over CaH₂, followed by distillation over molecular sieves (4 Å). Synthesis of Calone 1951® (1) by means of the microwave heating procedures has been discussed previously [8], and intermediates 10d and 10e were more efficiently prepared by the microwave methodology. Microwave experiments were performed in a prototype microwave applicator 'MDV2.4' and with a 0-700 W variable power microwave source (2m172)²). FC: Merck silica gel 60 (0.040–0.063 μm). CC = Column chromatography. TLC: Merck silica gel 60 F₂₅₄ (particle size 5–40 μm, layer thickness 0.2 mm on aluminium, 20 × 20 cm); visualization by UV light (254 nm). Semiprep. HPLC: Varian Prostar (model 210); Phenomenex column (250 × 10 mm, 5μ ODS (3)); MeCN/H₂O 33:67; 3.5 ml/min flow rate; UV/VIS detection. IR: Perkin-Elmer Spectrum-2000 Fourier-transform IR spectrophotometer. NMR: Bruker Avance-300 (300 MHz) spectrometer, Varian Gemini-200 (200 MHz) spectrometer, referenced to SiMe₄ with the solvent resonance as the internal standard (CHCl₃: δ(H) 7.26) unless otherwise stated. MS/GC: Hewlett-Packard 6890 GC with BPX-5 column/5973 mass-selective detector.

- 2. Calculations. Molecular-model calculations were performed by using density functional theory (DFT), as implemented in the Dmol³ program package. The PBE functional was used in combination with the DNP basis set (double numerical plus polarization). All calculations were all-electron and spin-restricted. Each model was charge-neutral. The geometry of each model was optimized, without constraints. Electron-density maps, HOMOs and LUMOs were calculated for the relaxed structures.
- 3. Calone 1951® (1) and Analogues 12–18: General Procedure A. 2,2'-[(4-Methyl-1,2-phenylene)bis(oxy)]bis[acetic Acid] Dimethyl Ester (10; R¹=4-Me). Conventional Method: Ground, ovendried K_2CO_3 (9.00 g, 65.12 mmol) and methyl bromoacetate (6.00 ml, 63.38 mmol) were added sequentially to a soln. of 9 (R¹=4-Me; 2.00 g, 16.11 mmol) in anh. DMF (100 ml). The soln. was heated to 120° for 2 h, then the hot mixture was poured into ice/ H_2O (200 ml) and extracted with CH_2Cl_2 (3 × 100 ml). The combined org. phase was washed with 5% aq. NaOH soln. (2 × 50 ml) followed by H_2O (2 × 50 ml), dried (MgSO₄), and concentrated. The obtained yellow oil was purified by vacuum CC (silica gel (short column), pentane, then AcOEt/pentane $1:9 \rightarrow 3:7$) followed by bulb-to-bulb distillation: $10 (R^1 = 4-Me; 3.37 g, 78\%)$. White solid. M.p. (pentane) $43-44^\circ$.

Microwave Procedure: The reagents were combined as described for the conventional procedure, and the soln. was irradiated for 4 min at 200 W under N_2 . Workup of the mixture was performed according to the conventional procedure: 10 ($R^1 = 4$ -Me; 3.60 g, 83%).

²⁾ Provided on trial by Warlock Engineering: www.warlock.com.au/chemreactor.htm.

Methyl 3,4-Dihydro-7-methyl-3-oxo-2H-1,5-benzodioxepine-2-carboxylate (11; R^1 =7-Me). Under an inert atmosphere, a soln. of 10 (3.72 g, 13.88 mmol) in anh. THF (35 ml) was added within 5 min to a stirred mixture of KO'Bu (3.11 g, 27.71 mmol) in anh. THF (35 ml) with cooling in an ice/H₂O bath. The soln. was then transferred to an oil bath at 70° and heated for 30 min before quenching in a stirred mixture of 0.2m HCl (200 ml) in ice (200 g). The acidic soln. was extracted with CH₂Cl₂ (3 × 100 ml) and the combined org. phase washed with H₂O (2 × 150 ml) and concentrated: 11 (R^1 =7-Me; 3.00 g, 92%). Viscous yellow-brown oil consisting of a mixture of regioisomers and requiring storage at -20° . After prolonged storage, purification by CC (silica gel, hexane/AcOEt 1:1) was necessary prior to use.

7-Methyl-2H-1,5-benzodioxepin-3(4H)-one (1). To a soln. of 11 (R¹ = 7-Me; 1.00 g, 4.24 mmol) in EtOH (10 ml) was added 2M HCl (50 ml), and the soln. was heated at 90° while stirring in an oil bath. After 2 h, the mixture was poured into ice/H₂O (100 ml) and extracted with CH₂Cl₂ (3 × 100 ml). The combined org. extract was washed with H₂O (1 × 200 ml), dried (MgSO₄), and concentrated and the obtained brown semi-solid subjected to bulb-to-bulb distillation (70°/0.5 Torr): 1 (0.59 g, 78%). White crystalline solid. M.p. (pentane) $38-39^{\circ}$ (lit.³) $38-40^{\circ}$).

4. Calone 1951° (1) and Analogues 12 – 18. General Procedure B. 1-Bromo-3-chloropropan-2-ol (23). A 48% HBr soln. (28.3 g) was added dropwise to neat epichlorohydrin (=2-(chloromethyl)oxirane; 22; 15.0 g, 162 mmol) within 2 h under vigorous stirring. The soln. was stirred for an additional 45 min, then mixed with Et₂O (150 ml) and washed with sat. NaHCO₃ soln. (3 × 50 ml). The org. phase was dried (MgSO₄) and concentrated and the residue distilled under vacuum (54 – 56°/5 Torr): 23 (20.0 g, 71%). Clear oil. IR (neat): 3380s (br.), 2961m, 2894m, 1625m, 1426m, 1382m, 1344m, 1295m, 1261m, 1225m, 1135m, 1088m, 1067m, 1039m, 14-NMR (300 MHz, CDCl₃): 4.03 (quint., J = 5.3, H – C(2)); 3.70 (d, J = 5.7, CH₂(3)); 3.56 (d, J = 5.3, CH₂(1)); 2.64 (m, OH). ¹³C-NMR (75 MHz, CDCl₃): 70.3 (C(2)); 46.4 (C(3)); 34.7 (C(1)). ¹H, ¹H-COSY (CDCl₃): 3.98 (H – C(2))/3.63 (H – C(3)); 3.98 (H – C(2))/3.51 (H – C(1)); 3.65 (H – C(1), H – C(3))/2.58 (OH). ¹H, ¹³C-HMQC (CDCl₃): H – C(2)/C(2) (70.5); CH₂(3)/C(3) (46.7); CH₂(1)/C(1) (34.9). EI-MS: 125 (96), 123 (100), 95 (6), 93 (6), 81 (18), 79 (50).

2-[2-Bromo-1-(chloromethyl)ethoxy]tetrahydro-2H-pyran (24). To 3,4-dihydro-2H-pyran (5.30 g, 63.00 mmol), 23 (10.0 g, 57.66 mmol) was added neat, followed by a cat. amount of P_2O_5 (0.10 g, 0.35 mmol). The soln. was warmed to 60° in a water bath for 1 h and then allowed to reach r.t. and stirred for an additional hour. The resulting mixture was dissolved in Et_2O (150 ml) and washed with sat. NaHCO3 soln. (3 × 50 ml), the org. phase dried (MgSO4) and concentrated, and the residue distilled under vacuum (101 – 102°/50 Torr): 24 (13.5 g, 91%). Clear oil. IR (neat): 2940m, 2868w, 2853w, 1437w, 1395w, 1382w, 1384w, 1323w, 1284w, 1261w, 1199m, 1178w, 1148w, 1122m, 1075m, 1054s, 1029s. ¹H-NMR (300 MHz, CDCl3): 4.78–4.77 (m, H–C(2)); 4.06–3.98 (m, ClCH2CH); 3.92–3.48 (m, CH2(6)); 3.83–3.48 (m, ClCH2, BrCH2); 1.82–1.50 (m, CH2(3), CH2(4), CH2(5)). 13 C-NMR (75 MHz, CDCl3): 98.8 (C(2)); 75.5 (CH); 62.7 (C(6)); 44.2 (ClCH2); 31.9 (BrCH2); 30.5 (C(3)); 25.2 (C(5)); 19.2 (C(4)). 14 H₃C-HMQC (CDCl3): CH2(2)/C(2) (99.1); H–C(1)/C(1) (75.6); CH2(6)/C(6) (62.7); Ha-C(1)/ClCH2 (44.7); H_b-C(1)/ClCH2 (44.6); CH2(2)/C(2) (32.4); CH2(3)/C(3) (30.3); CH2(5)/C(5) (25.3); CH2(4)/C(4) (19.2). EI-MS: 257 (10, M^+), 255 (8), 209 (2), 207 (2), 203 (2), 202 (2), 201 (2), 200 (3), 157 (16), 155 (12), 137 (4), 135 (4), 121 (5), 101 (4), 95 (3), 93 (2), 85 (100), 77 (6), 75 (18), 67 (7), 56 (20).

3,4-Dihydro-7-methyl-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-1,5-benzodioxepine (26; R¹=7-Me). Ground oven-dried K_2CO_3 (2.22 g, 16.06 mmol) was added in one portion to a soln. of 9 (0.50 g, 4.03 mmol) in DMF, and the soln. was heated to 100° . Then a soln. of 24 (0.70 g, 2.72 mmol) in DMF was added dropwise over 30 min and the soln. stirred at 120° for 24 h. The hot soln. was poured into a stirred ice/ H_2O (200 ml) mixture and saturated with NaCl before extraction with CH_2Cl_2 (3 × 75 ml). The combined org. phase was washed with H_2O (2 × 100 ml), dried (MgSO₄), and concentrated: 26 (0.98 g, 92%). Deep brown viscous oil. IR (neat): 3436 (br.), 3031w, 2942s, 2871s, 2738w, 2660w, 2578w, 2251w, 2227w, 2052w, 2017w, 1871w, 1724w, 1614w, 1579m, 1505s, 1455m, 1442m, 1415w, 1381w, 1355m, 1305s, 1262s, 1201s, 1185m, 1150m, 1128s, 1075s, 1034s. ¹H-NMR (300 MHz, CDCl₃): 6.91–6.66 (m, H–C(6), H–C(8), H–C(9)); 4.74 (t, t = 3.5, H–C(2')); 4.42–4.00 (t c, t C,

³⁾ Material safety data sheet of 7-methyl-3,4-dihydro-2H-1,6-benzodioxepine-3-one (Calone 1951®); Ganone Agan Aroma and Fine Chemicals Ltd.

 $\begin{array}{l} (75 \ MHz, CDCl_3): \ 149.6 \ (C(9a)); \ 147.8 \ (C(5a)); \ 132.7 \ (C(7)); \ 124.4 \ (C(8)); \ 123.4 \ (C(9)); \ 121.2 \ (C(6)); \\ 98.4 \ (C(2')); \ 74.4 \ (C(3)); \ 73.0 \ (C(4)); \ 72.1 \ (C(2)); \ 62.7 \ (C(6')); \ 30.7 \ (C(3')); \ 25.3 \ (C(5')); \ 20.5 \ (Me); \\ 19.41 \ (C(4')). \ ^1H, ^1H-COSY \ (CDCl_3): \ 4.66 \ (H-C(2'))/1.61 \ (CH_2(3')); \ 3.81 \ (H_b-C(6'))/3.45 \ (H_a-C(6')); \\ 3.81 \ (H-C(6'))/1.49 \ (CH_2(5')). \ ^1H, ^13C-HMQC \ (CDCl_3): \ H-C(3)/C(3) \ (74.5); \ CH_2(4)/C(4) \ (73.4); \\ CH_2(2)/C(2) \ (72.1); \ CH_2(6')/C(6') \ (62.7). \ EI-MS: \ 264 \ (77, M^+), \ 236 \ (3), \ 208 \ (4), \ 206 \ (2), \ 180 \ (20), \ 162 \ (22), \ 149 \ (7), \ 135 \ (35), \ 11 \ (133), \ 124 \ (8), \ 123 \ (9), \ 105 \ (7), \ 99 \ (6), \ 91 \ (7), \ 85 \ (100), \ 77 \ (10), \ 67.1 \ (19), \ 57 \ (23), \ 55 \ (10). \end{array}$

3,4-Dihydro-7-methyl-2H-1,5-benzodioxepin-3-ol (27). Vanadium pentoxide (14 mg, 0.11 mmol) was added to a 30% H_2O_2 soln. (0.90 ml), and the mixture was stirred at 0° for 10 min. The catalyst mixture was then added to a soln. of 26 (0.50 g, 1.89 mmol) in MeCN (20 ml) and heated at 70° for 15-20 min, or until the soln. ceased effervescing and became dark green. The MeCN was evaporated and the crude material solubilized in 2M NaHCO₃ (50 ml). Extraction of the product was achieved with CH₂Cl₂ (3× 40 ml), and the org. phase washed with sat. NaHCO₃ soln. $(1 \times 60 \text{ ml})$ and concentrated: opaque resin which solidified upon cooling to give 27 (0.26 g, 76%). White solid. M.p. (heptane) 56-58°. IR (KBr): 3210 (br.), 3083m, 2988m, 2961m, 2922m, 2866m, 2714w, 1611w, 1578m, 1509s, 1443m, 1414w, 1383w, 1360w, 1344s, 1300s, 1290s, 1275s, 1261s, 1202m, 1151m, 1137s, 1116m, 1103w, 1042s. ¹H-NMR (300 MHz, $CDCl_3$): 6.89 (d, J = 8.1, H - C(9)); 6.80 (s, H - C(6)); 6.78 (d, J = 8.1, H - C(8)); 4.35 – 4.25 (m, $H_a - C(2)$, $H_a-C(4)$; 4.08-3.99 (m, $H_b-C(4)$, H-C(3), $H_b-C(2)$); 2.78 (s, OH); 2.24 (s, Me). ¹³C-NMR (75 MHz, CDCl₃): 150.2 (C(9a)); 148.3 (C(5a)); 133.0 (C(7)); 123.7 (C(8)); 121.5 (C(9)); 120.7 (C(6)); 74.6 (C(2)); 74.4 (C(4)); 69.1 (C(3)); 20.3 (Me). ${}^{1}H, {}^{1}H-COSY$ (CDCl₃): 4.30 (H_a-C(2))/4.03 $(H_b-C(2))$; 4.30 $(H_a-C(4))/4.03$ $(H_b-C(4))$. 1H , ${}^{13}C$ -HMQC $(CDCl_3)$: $CH_2(2)/C(2)$ (74.5); $CH_2(4)/C(4)$ C(4) (74.4); H-C(3)/C(3) (69.2). EI-MS: 180 (100, M^+), 161 (2), 149 (11), 135 (51), 123 (28), 109 (11), 107 (5), 106 (5), 95 (8), 94 (9), 91 (10), 78 (13), 77 (14), 66 (10), 65 (6) 57 (2), 51 (6).

7-Methyl-2H-1,5-benzodioxepin-3(4H)-one (1). To a 4% aq. KOH soln. (20 ml) and KMnO₄ (1.75 g, 11.07 mmol) at $0-5^{\circ}$ was added 27 (0.40 g, 2.22 mmol) in one portion. The soln. was stirred for 2.5 h, then the MnO₂ was removed by vacuum filtration and the filter cake washed with Et₂O (2 × 20 ml). The filtrate was acidified by the dropwise addition of conc. HCl soln. to pH 2, and then extracted with Et₂O (3 × 30 ml). The combined org. phase was washed with sat. NaHCO₃ soln. (1 × 50 ml), dried (MgSO₄), and concentrated and the obtained yellow resin purified by bulb-to-bulb distillation (70°/0.5 Torr): 1 (0.35 g, 87%). Crystalline white solid. M.p. (pentane) 38–39°. IR (KBr): 3448w, 3057w, 2991w, 2920w, 2859w, 1742s, 1614w, 1582w, 1506s, 1466w, 1435m, 1414w, 1303s, 1265s, 1207m, 1150m, 1119m, 1053s. ¹H-NMR (300 MHz, CDCl₃): 6.90–6.75 (m, H–C(6), H–C(8), H–C(9)); 4.70 (s, CH₂(4)); 4.67 (s, CH₂(2)); 2.27 (s, Me). ¹³C-NMR (75 MHz, CDCl₃): 204.8 (C(3)); 147.9 (C(9a)); 146.1 (C(5a)); 133.8 (C(7)); 124.3 (C(8)); 121.2 (C(9)); 120.6 (C(6)); 75.8 (C(2)); 75.5 (C(4)); 20.5 (Me). EI-MS: 178 (100, M^+), 150 (2), 149 (3), 135 (24), 122 (6), 108 (8), 94 (89), 91 (17), 89 (12), 77 (17), 66 (74), 63 (14), 51 (18), 39 (21).

5. Benzodioxepinone 12 by Procedure A. 2,2'-[(3-Methyl-1,2-phenylene)bis(oxy)]bis[acetic Acid] Dimethyl Ester (10a). As described in Exper. 3 (conventional method), with 9a (1.00 g, 8.05 mmol). Purification involved bulb-to-bulb distillation (160°/0.4 Torr) performed twice: 10a (1.99 g, 92%). Clear pale yellow oil. The microwave procedure provided 10a in 36% yield. IR (neat): 3002m, 2955s, 2853m, 2112w, 1774s, 1749s, 1646w, 1604m, 1587m, 1488s, 1440s, 1380m, 1304m, 1262m, 1183m, 1114m, 1070m, 1027m. ¹H-NMR (300 MHz, CDCl₃)¹): 6.89 – 6.60 (m, H – C(4'), H – C(5'), H – C(6')); 4.69 (s, CH₂(1'')); 4.60 (s, CH₂(2)); 3.74 (s, Me(3")O); 3.72 (s, Me(3)O); 2.28 (s, arom. Me). ¹³C-NMR (75 MHz, CDCl₃)¹): 169.9 (C(2")O); 168.9 (C(1)O); 149.8 (C(1')); 145.9 (C(2')); 132.3 (C(3')); 124.1 (C(4')); 123.7 (C(5')); 111.5 (C(6')); 69.0 CH₂(1'')); 65.6 (CH₂(2)); 51.6 (Me(3)O, Me(3")O); 15.8 (arom. Me). EI-MS: 268 (16, M+), 209 (5), 207 (3), 195 (11), 176 (11), 163 (12), 149 (15), 135 (75), 121 (3), 109 (17), 91 (19), 77 (21), 65 (17), 63 (5), 59 (9), 51 (10), 45 (100).

Methyl 3,4-Dihydro-6-methyl-3-oxo- and 3,4-Dihydro-9-methyl-3-oxo-2H-1,5-benzodioxepine-2-carboxylate (**11a**). As described in *Exper. 3*, with **10a** (1.00 g, 3.73 mmol): **11a** (0.72 g, 82%). Caramel-colored oil. IR (neat): 3474m, 2956s, 2572w, 1742s, 1677m, 1632w, 1594m, 1477s, 1440s, 1337m, 1274s, 1250m, 1217s, 1202s, 1100s, 1071m, 1045m. 1 H-NMR (300 MHz, CDCl₃): *Regioisomer 1*: 6.93-6.83 (m, H-C(7), H-C(8), H-C(9)); 5.36 (s, H-C(2)); 4.81 (d, J = 16.7, H_a-C(4)); 4.62 (d, J = 16.7, H_b-C(4)); 3.86 (s, MeO); 2.22 (s, arom. Me); *Regioisomer 2*: 6.93-6.83 (m, H-C(7), H-C(8), H-C(9)); 5.37 (s,

H–C(4)); 4.87 (d, J = 16.7, H_a–C(2)); 4.56 (d, J = 16.7, H_b–C(2)); 3.85 (s, MeO); 2.28 (s, arom. Me). ¹³C-NMR (75 MHz, CDCl₃): Regioisomer 1: 199.7 (C(3)); 165.4 (CO₂); 147.1 (C(9a)); 145.9 (C(5a)); 130.7 (C(6)); 125.2 (C(7)); 124.0 (C(8)); 118.1 (C(9)); 81.8 (C(2)); 76.4 (C(4)); 52.9 (MeO); 15.9 (arom. Me); Regioisomer 2: 199.6 (C(3)); 165.4 (CO₂); 147.3 (C(9a)); 145.4 (C(5a)); 130.0 (C(6)); 126.4 (C(7)); 122.8 (C(8)); 118.8 (C(9)); 81.8 (C(4)); 76.4 (C(2)); 58.4 (MeO); 18.2 (arom. Me). EI-MS: 236 (100, M^+), 204 (4), 177 (22), 149 (66), 135 (48), 121 (22), 105 (5), 94 (47), 77 (28), 66 (34), 51 (13), 39 (18).

6-Methyl-2H-1,5-benzodioxepin-3(4H)-one (12). As described in Exper. 3, with 11a (1.02 g, 4.32 mmol). The obtained brown oily 12 (0.56 g, 72%) which was bulb-to-bulb distilled at 50–60°/0.2 Torr: 12 (0.11 g, 14%). Clear oil. IR (KBr): 3024w, 2964w, 2939w, 2906w, 2839w, 1790m, 1738s, 1613w, 1592m, 1492m, 1475s, 1440m, 1419w, 1380w, 1324m, 1296m, 1274s, 1258s, 1205m, 1191m, 1101s, 1080m.

1H-NMR (300 MHz, CDCl₃): 6.86 (br. s, H–C(7), H–C(8), H–C(9)); 4.74 (s, CH₂(4)); 4.70 (s, CH₂(2)); 2.23 (s, Me).

13C-NMR (75 MHz, CDCl₃): 205.1 (C(3)); 148.4 (C(9a)); 146.9 (C(5a)); 130.4 (C(6)); 125.4 (C(7)); 123.0 (C(8)); 118.6 (C(9)); 75.6 (C(4)); 75.5 (C(2)); 16.1 (Me). EI-MS: 178 (100, M⁺), 150 (5), 135 (32), 122 (2), 108 (5), 94 (82), 77 (17), 66 (77), 51 (17), 39 (19).

6. Benzodioxepinone 12 by Procedure B. 3,4-Dihydro-6-methyl-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-1,5-benzodioxepine (26a). As described in Exper. 4, with 9a (0.50 g, 4.03 mmol): 26a (0.95 g, 90%). Brown viscous oil. IR (neat): 3452 (br.), 3075w, 3017w, 2943s, 2870s, 2739w, 1906w, 1835w, 1724w, 1649w, 1624w, 1593m, 1477s, 1455m, 1440m, 1402w, 1379w, 1355m, 1302s, 1272s, 1248m, 1194s, 1159m, 1128s, 1084s, 1034s. 1 H-NMR (300 MHz, CDCl₃): 6.77 (s, H-C(7), H-C(8), H-C(9)); 4.75 ('t', H-C(2')); 4.45-4.23 (m, CH₂(2), H-C(3), CH₂(4)); 3.95-3.87 (m, H_a-C(6')); 3.57-3.50 (m, H_b-C(6')); 2.21 (s, Me); 1.90-1.51 (m, CH₂(3'), CH₂(4'), CH₂(5')). 13 C-NMR (75 MHz, CDCl₃): 150.0 (C(9a)); 148.5 (C(5a)); 130.0 (C(6)); 124.4 (C(7)); 122.0 (C(8)); 118.4 (C(9)); 98.4 (C(2')); 74.4 (C(3)); 72.9 (C(2)); 71.9 (C(4)); 62.7 (C(6')); 30.7 (C(3')); 25.3 (C(5')); 19.5 (C(4')); 16.2 (Me). EI-MS: 264 (58, M^+), 236 (6), 208 (3), 206 (1), 180 (14), 162 (21), 149 (8), 135 (38), 124 (5), 123 (6), 105 (5), 99 (5), 94 (5), 91 (7), 85 (100), 77 (10), 67 (18), 57 (22), 55 (8).

3,4-Dihydro-6-methyl-2H-1,5-benzodioxepin-3-ol (**27a**). As described in *Exper. 4*, with **26a** (0.70 g, 2.65 mmol): **27a** (0.41 g, 85%). Clear oil which solidified to a white solid upon cooling. M.p. (heptane) 34–38°. IR (KBr): 3265 (br.), 2964*m*, 2920*m*, 2860*w*, 1589*w*, 1528*w*, 1477*s*, 1438*m*, 1377*w*, 1343*w*, 1292*s*, 1278*s*, 1247*m*, 1196*s*, 1161*w*, 1138*s*, 1087*s*, 1045*s*, 1011*m*. ¹H-NMR (300 MHz, CDCl₃): 6.86 (*s*, H–C(7), H–C(8), H–C(9)); 4.35–4.25 (*m*, H–C(3)); 4.08–3.99 (*m*, CH₂(2), CH₂(4)); 2.60 (br. *s*, OH); 2.24 (*s*, Me). ¹³C-NMR (75 MHz, CDCl₃): 151.4 (C(9a)); 149.7 (C(5a)); 131.0 (C(6)); 125.5 (C(7)); 123.2 (C(8)); 119.2 (C(9)); 74.8 (C(2)); 74.4 (C(4)); 69.7 (C(3)); 16.1 (Me). EI-MS: 180 (100, *M*⁺), 161 (3), 149 (16), 136 (22), 135 (62), 124 (21), 123 (24), 109 (12), 107 (4), 105 (5), 95 (5), 94 (9), 91 (12), 89 (2), 78 (11), 77 (16), 66 (11), 57 (3), 52 (5), 51 (7).

6-Methyl-2H-1,5-benzodioxepin-3(4H)-one (12). As described in Exper. 4, with 27a (0.40 g, 2.22 mmol). The obtained yellow resin was purified by bulb-to-bulb distillation at $50-60^{\circ}/0.2$ Torr: 12 (0.22 g, 56%). Clear oil.

7. Benzodioxepinone **13** by Procedure A. 2,2'-[(3-Methoxy-1,2-phenylene)bis(oxy)]bis[acetic Acid] Dimethyl Ester (**10b**). As described in Exper. 3 (conventional method), with **9b** (2.51 g, 17.9 mmol). Purification involved bulb-to-bulb distillation (190°/0.5 Torr) performed twice: **10b** (3.35 g, 66%). Clear oil. The microwave procedure provided **10b** in 45% yield. IR (KBr): 2954w, 2918w, 2843w, 1757s, 1602m, 1499m, 1479s, 1445m, 1397w, 1387w, 1306m, 1259m, 1217s, 1182m, 1125s, 1093m, 1069m. ¹H-NMR (300 MHz, CDCl₃)¹): 6.94 – 6.45 (m, H – C(4'), H – C(5'), H – C(6')); 4.65 (s, CH₂(1'')); 4.64 (s, CH₂(2)); 3.79 (s, arom. MeO); 3.75 (s, Me(3")O); 3.72 (s, Me(3)O). ¹³C-NMR (75 MHz, CDCl₃)¹): 169.4 (C(1)O); 169.1 (C(2")O); 153.3 (C(1')); 151.2 (C(3')); 137.3 (C(2')); 123.8 (C(5')); 107.8 (C(4')); 106.7 (C(6')); 69.5 (CH₂(2)); 66.6 (CH₂(1"')); 56.1 (arom. MeO); 51.6 (Me(3)O, Me(3")O). EI-MS: 284 (85, M+), 252 (2), 211 (64), 192 (12), 165 (20), 153 (87), 137 (9), 125 (40), 107 (29), 95 (32), 77 (15), 65 (12), 51 (16), 45 (100).

Methyl 3,4-Dihydro-6-methoxy-3-oxo- and 3,4-Dihydro-9-methoxy-3-oxo-2H-1,5-benzodioxepine-2-carboxylate (11b). As described in *Exper. 3*, with 10b (3.32 g, 11.69 mmol). Trituration with EtOH provided 11b (2.50 g, 85%). White crystalline solid. M.p. (EtOH) 117 – 119°. IR (KBr): 2962m, 2897w, 2847w, 1767s, 1741s, 1596m, 1486s, 1473s, 1441m, 1435m, 1416w, 1345m, 1319m, 1277s, 1259m, 1246m, 1214s, 1204s, 1176s, 1105s, 1080s, 1068m, 1034m. ¹H-NMR (300 MHz, CDCl₃): 6.92 – 6.60 (m, H – C(7),

H-C(8), H-C(9)); 5.36 (s, H-C(2)); 4.75 (d, J = 16.6, H $_a$ -C(4)); 4.54 (d, J = 16.6, H $_b$ -C(4)); 3.88 (s, arom. MeO); 3.84 (s, COOMe). 13 C-NMR (75 MHz, CDCl $_3$): 198.9 (C(3)); 164.9 (CO $_2$); 150.9 (C(9a)); 148.3 (C(6)); 137.0 (C(5a)); 122.6 (C(8)); 113.4 (C(7)); 108.0 (C(9)); 82.4 (C(2)); 76.9 (C(4)); 56.4 (arom. MeO); 52.9 (COOMe). EI-MS: 252 (97, M⁺), 209 (16), 193 (15), 165 (28), 151 (30), 137 (9), 110 (46), 95 (100), 79 (19), 65 (16), 51 (31), 39 (53).

6-Methoxy-2H-1,5-benzodioxepin-3(4H)-one (13). As described in Exper. 3, with 11b (1.10 g, 4.36 mmol) in a minimum amount of EtOH/CH₂Cl₂ 2:1 for 2.5 h. Aq. workup involved neutralization with NaHCO₃ prior to extraction: 13 (0.81 g, 95%). White crystals. M.p. (pentane) $130-132^{\circ}$. IR (KBr): 3030w, 2962m, 2897w, 2847w, 1767s, 1740s, 1596m, 1486s, 1474s, 1441m, 1435m, 1415w, 1345w, 1319m, 1277m, 1259m, 1247m, 1214s, 1204s, 1176m, 1105s, 1080m, 1068m, 1034w. ¹H-NMR (300 MHz, CDCl₃): 6.94-6.62 (m, H-C(7), H-C(8), H-C(9)); 4.87 (s, $CH_2(4)$); 4.73 (s, $CH_2(2)$); 3.87 (s, MeO). ¹³C-NMR (75 MHz, CDCl₃): 204.2 (C(3)); 151.5 (C(6)); 149.4 (C(9a)); 138.4 (C(5a)); 122.9 (C(8)); 113.1 (C(7)); 106.7 (C(9)); 76.2 (C(2)); 75.6 (C(4)); 56.3 (MeO). EI-MS: 194 (100, M^+), 166 (2), 151 (63), 137 (3), 123 (5), 110 (32), 107 (18), 95 (75), 79 (10), 77 (5), 65 (10), 51 (15), 39 (21). HR-ESI-MS: 195.0650 ($C_{10}H_{10}O_4^+$, $[M+1]^+$; calc. 195.0657).

8. Benzodioxepinone 13 by Procedure B. 3,4-Dihydro-6-methoxy-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-1,5-benzodioxepine (26b). As described in Exper. 4, with 9b (0.50 g, 3.57 mmol): 26b (0.91 g, 91%). Brown viscous oil. IR (neat): 3436 (br.), 2944s, 2873m, 2853m, 2744w, 2251w, 1673m, 1652m, 1598m, 1588m, 1489s, 1476s, 1456m, 1441m, 1387w, 1356w, 1311m, 1273m, 1248m, 1202s, 1183w, 1162w, 1119s, 1103s, 1077s, 1093s, 1024s. ¹H-NMR (300 MHz, CDCl₃): 6.87 – 6.78 (m, H-C(8)); 6.62 – 6.51 (m, H-C(7), H-C(9)); 4.96 – 4.43 (m, H $_a-C(2)$, H $_a-C(4)$); 4.56 – 4.04 (m, H $_b-C(2)$, H-C(3), H $_b-C(4)$); 5.10 (t, J = 3.5, H-C(2')); 3.84 (s, MeO); 3.56 – 3.49 (m, CH $_2(6')$); 1.87 – 1.53 (m, CH $_2(3')$, CH $_2(4')$, CH $_2(5')$). 13 C-NMR (75 MHz, CDCl $_3$): 151.2 (C(6)); 143.9 (C(9a)); 139.8 (C(5a)); 120.6 (C(8)); 113.2 (C(9)); 110.4 (C(7)); 98.0 (C(2')); 74.1 (C(3)); 72.3 (C(2)); 70.2 (C(4)); 62.9 (C(6')); 56.1 (MeO); 30.7 (C(3')); 25.3 (C(5')); 23.2 (C(4')). EI-MS: 280 (50, M^+), 252 (13), 224 (4), 196 (29), 178 (18), 165 (7), 151 (36), 149 (11), 140 (12), 125 (5), 107 (14), 95 (18), 85 (100), 77 (5), 67 (20), 57 (29), 55 (15), 51 (6).

3,4-Dihydro-6-methoxy-2H-1,5-benzodioxepin-3-ol (**27b**). As described in Exper. 4, with **26b** (0.77 g, 2.75 mmol): **27b** (0.41 g, 76%). Yellow resin. IR (neat): 3459 (br.), 3099w, 2941m, 2840w, 1721m, 1646w, 1598m, 1588m, 1489s, 1475s, 1442m, 1361w, 1321w, 1304w, 1272s, 1249s, 1207m, 1183m, 1092s, 1076s, 1029m. 1 H-NMR (300 MHz, CDCl₃): 6.89 (t, t = 8.3, H-C(8)); 6.67-6.61 (t , H-C(7), H-C(9)); 4.41-4.26 (t , H_a-C(4), H_a-C(2)); 4.16-4.09 (t , H_b-C(4), H-C(3), H_b-C(2)); 3.86 (t , MeO); 2.01 (br. t , OH). 1 3C-NMR (75 MHz, CDCl₃): 152.3 (C(6)); 151.9 (C(9a)); 140.8 (C(5a)); 122.9 (C(8)); 113.7 (C(9)); 107.0 (C(7)); 75.2 (C(2)); 74.9 (C(4)); 69.7 (C(3)); 56.3 (MeO). EI-MS: 196 (100, t , 177 (1), 165 (9), 151 (38), 140 (25), 125 (15), 110 (11), 107 (26), 95 (21), 93 (17), 81 (4), 77 (5), 68 (2), 66 (2), 6.3 (3), 57 (4), 55 (4), 51 (10).

6-Methoxy-2H-1,5-benzodioxepin-3(4H)-one (13). As described in Exper. 4, with 27b (0.40 g, 2.04 mmol). The obtained yellow resin was bulb-to-bulb distilled at 65°/0.2 Torr: 13 (0.34 g, 87%). Clear crystals.

9. Benzodioxepinone **14** by Procedure A. 2,2'-[1,2-Phenylenebis(oxy)]bis[acetic Acid] Dimethyl Ester (**10c**). As described in Exper. 3 (conventional method), with **9c** (0.50 g, 4.54 mmol). Bulb-to-bulb distillation (180°/0.05 Torr) performed twice afforded **10c** (1.01 g, 88%). White solid. M.p. (pentane) 44–48°. The microwave procedure provided **10c** in 48% yield. IR (KBr): 3012w, 2961m, 1752s, 1708w, 1596m, 1507s, 1456m, 1434m, 1372m, 1334m, 1275m, 1233s, 1197s, 1180s, 1163m, 1138s, 1088w, 1068m, 1020w.

¹H-NMR (300 MHz, CDCl₃)¹): 6.94–6.86 (m, H–C(3'), H–C(4'), H–C(5'), H–C(6')); 4.71 (s, CH₂(1"), CH₂(2)); 3.77 (s, Me(3)O, Me(3")O).

¹G-NMR (75 MHz, CDCl₃)¹): 169.3 (C(1)O, C(2")O); 147.9 (C(1'), C(2')); 122.6 (C(4'), C(5')); 115.3 (C(3'), C(6')); 66.5 (CH₂(1"), CH₂(2)); 52.1 (Me(3)O, Me(3")O). EI-MS: 254 (32, M⁺), 222 (2), 193 (10), 162 (33), 149 (3), 135 (14), 121 (29), 107 (2), 95 (10), 77 (16), 63 (8), 52 (10), 45 (100).

*Methyl 3,4-Dihydro-3-oxo-*2H-*1,5-benzodioxepine-2-carboxylate* (**11c**). As described in *Exper. 3*, with **10c** (3.65 g, 14.37 mmol): **11c** (2.78 g, 87%). Pale brown oil. IR (neat): 3472*m*, 2956*m*, 1746*s*, 1682*m*, 1634*w*, 1594*m*, 1494*s*, 1456*m*, 1439*s*, 1396*w*, 1338*m*, 1273*s*, 1250*m*, 1228*s*, 1157*m*, 1106*m*, 1081*m*, 1025*m*. ¹H-NMR (300 MHz, CDCl₃): 7.11–6.95 (*m*, H–C(6), H–C(7), H–C(8), H–C(9)); 5.36 (*s*, H–C(2));

4.82 $(d, J=16.8, H_a-C(4))$; 4.60 $(d, J=16.8, H_b-C(4))$; 3.85 (s, MeO). ¹³C-NMR (75 MHz, CDCl₃): 199.3 (C(3)); 165.2 (CO₂); 147.2 (C(5a)); 147.0 (C(9a)); 125.0 (C(7)); 123.6 (C(9)); 121.4 (C(8)); 120.5 (C(6)); 81.8 (C(2)); 76.8 (C(4)); 52.9 (MeO). EI-MS: 222 (99, M^+), 190 (6), 163 (17), 146 (3), 135 (100), 121 (61), 107 (24), 92 (4) 91 (4), 80 (52), 77 (38), 69 (11), 65 (26), 63 (26), 52 (41), 43 (30).

 $2\text{H-}1,5\text{-}Benzodioxepin-}3(4\text{H})\text{-}one$ (14). As described in *Exper. 3*, with 11c (0.65 g, 2.93 mmol) in EtOH/CH₂Cl₂ 2:1. The obtained yellow-brown semi-solid 14 (0.45 g, 94%) was triturated with CH₂Cl₂: 14 (0.05 g, 10%). White crystalline solid. M.p. (CH₂Cl₂) 64–65°. IR (KBr): 3408w, 2957m, 1741s, 1631m, 1586w, 1495s, 1459w, 1303m, 1263s, 1181w, 1153w, 1097m, 1052s, 1025m. $^{1}\text{H-NMR}$ (300 MHz, CDCl₃): 6.90 (s, H–C(6), H–C(7), H–C(8), H–C(9)); 4.08 (s, CH₂(2), CH₂(4)). $^{13}\text{C-NMR}$ (75 MHz, (CD₃)₂CO): 204.3 (C(3)); 152.2 (C(5a), C(9a)); 124.9 (C(7), C(8)); 122.7 (C(6), C(9)); 79.4 (C(2), C(4)). EI-MS: 164 (100, M^+), 136 (2), 121 (27), 108 (10), 94 (3), 80 (75), 63 (13), 52 (27), 32 (22).

10. Benzodioxepinone **18** by Procedure B. 3,4-Dihydro-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-1,5-benzodioxepine (**26c**). As described in Exper. 4, with **9c** (0.50 g, 4.54 mmol): **26c** (0.99 g, 87%). Yellowbrown viscous oil. IR (neat): 3436 (br.), 3069w, 3036w, 2942s, 2871m, 2852m, 2741w, 2661w, 2223w, 1932w, 1737w, 1600w, 1583m, 1492s, 1464m, 1455m, 1442w, 1403w, 1379w, 1356w, 1303s, 1258s, 1248s, 1202w, 1181w, 1160m, 1122m, 1104m, 1074s, 1043s, 1022s. ¹H-NMR (300 MHz, CDCl₃): 7.01 – 6.86 (m, H – C(6), H – C(7), H – C(8), H – C(9)); 4.74 (t, J = 3.6, H – C(2')); 4.45 – 4.10 (m, CH₂(2), H – C(3), CH₂(4'); 3.94 – 3.87 (m, H_a – C(6')); 3.57 – 3.50 (m, H_b – C(6')); 1.89 – 1.50 (m, CH₂(3'), CH₂(4'), CH₂(5')). ¹³C-NMR (75 MHz, (CD₃)₂CO): 150.0 (C(5a), C(9a)); 124.0 (C(8)); 122.9 (C(7)); 121.5 (C(6)); 120.8 (C(9)); 98.5 (C(2')); 74.4 (C(3)); 73.1 (C(2)); 72.0 (C(4)); 62.7 (C(6')); 30.8 (C(3')); 25.3 (C(5')); 19.4 (C(4')). EI-MS: 250 (43, M⁺), 222 (2), 194 (2), 166 (18), 148 (23), 135 (7), 121 (37), 109 (5), 107 (1), 103 (1), 99 (5), 91 (4), 85 (100), 77 (7), 67.1 (15), 57 (18), 55 (11).

3,4-Dihydro-2H-1,5-benzodioxepin-3-ol (27c). As described in Exper. 4, with 26c (0.82 g, 3.28 mmol). The obtained clear oil partially solidified to a white semi-solid upon cooling. Evaporative crystallization from Et₂O/pentane gave 27c (0.49 g, 91%). Clear crystals. M.p. (pentane) $48-50^{\circ}$. IR (KBr): 3266 (br.), 2964w, 2917w, 2856w, 1594w, 1579w, 1523w, 1494m, 1455w, 1402w, 1375w, 1261s, 1182w, 1132m, 1098m, 1040s, 1019s. ¹H-NMR (300 MHz, CDCl₃): 7.04-6.94 (m, H-C(6), H-C(7), H-C(8), H-C(9)); 4.31-4.26 (m, H_a-C(4), H_a-C(2)); 4.12-4.07 (m, H_b-C(4), H-C(3), H_b-C(2)); 2.61 (br. s, OH). ¹³C-NMR (75 MHz, CDCl₃): 151.1 (C(5a), C(9a)); 124.0 (C(7), C(8)); 121.6 (C(6), C(9)); 74.8 (C(2), C(4)); 69.7 (C(3)). EI-MS: 166 (100, M^+), 147 (3), 135 (19), 122 (26), 121 (77), 110 (30), 109 (12), 95 (15), 81 (9), 77 (14), 63 (13), 52 (9).

2H-1,5-Benzodioxepin-3(4H)-one (14). As described in Exper. 4, with 27c (0.39 g, 2.35 mmol). The obtained pale yellow oil was purified by CC (silica gel (short column), CH_2Cl_2): 14 (0.23 g, 60%). White crystalline solid. M.p. (CH_2Cl_2) 64 -65° .

11. Benzodioxepinone **15** by Procedure A. 2,2'-[(4-Nitro-1,2-phenylenebis(oxy)]bis[acetic Acid] Dimethyl Ester (**10d**). As described in Exper. 3 (microwave procedure), with **9d** (0.50 g, 3.22 mmol). Purification by vacuum CC (silica gel (short column), pentane, then AcOEt/pentane $1:9 \rightarrow 3:7$ gradient), followed by bulb-to-bulb distillation (215°/0.1 Torr) provided **10d** (0.71 g, 74%). Bright yellow solid. The conventional method provided **10d** in 53% yield. M.p. (pentane) $84-86^{\circ}$. IR (KBr): 3087w, 2960w, 1757s, 1590w, 1520m, 1505m, 1437w, 1373w, 1347m, 1309w, 1286m, 1242w, 1207s, 1174m, 1145w, 1109m, 1085m, 1062m. ¹H-NMR (300 MHz, CDCl₃)¹): 7.93-6.85 (m, H-C(3'), H-C(5'), H-C(6')); 4.83 (s, CH₂(2)); 4.80 (s, CH₂(1'')); 3.82 (s, Me(3)O); 3.81 (s, Me(3'')O). ¹³C-NMR (75 MHz, CDCl₃)¹): 168.3 (C(1)O); 168.1 (C(2'')O); 153.0 (C(1')); 147.4 (C(2')); 142.2 (C(4')); 118.7 (C(6')); 113.0 (C(5')); 109.8 (C(3')); 66.1 (CH₂(2)); 66.0 (CH₂(1'')); 52.5 (Me(3)O); 52.4 (Me(3'')O). EI-MS: 299 (22, M^+), 267 (10), 240 (10), 207 (5), 194 (8), 180 (15), 166 (16), 151 (3), 135 (5), 120 (6), 107 (5), 92 (3), 79 (17), 63 (9), 51 (13), 45 (100), 30 (3).

Methyl 3,4-Dihydro-7-nitro-3-oxo- and 3,4-Dihydro-8-nitro-3-oxo-2H-1,5-benzodioxepine-2-carboxylate (11d). As described in *Exper. 3*, with 10d (0.71 g, 2.37 mmol): crude 11d (0.43 g, 68%). Bright red/yellow oil as a complex mixture. IR (neat): 3413*m*, 2924*s*, 2725*w*, 2255*w*, 2128*w*, 1740*m*, 1645*m*, 1590*w*, 1519*m*, 1460*s*, 1376*m*, 1342*m*, 1278*m*, 1051*s*, 1026*s*, 1005*s*. EI-MS: 267 (100, *M*⁺), 235 (11), 207 (22), 194 (6), 180 (8), 166 (29), 162 (10), 149 (8), 134 (66), 119 (21), 107 (24), 101 (7), 92 (16), 79 (53), 77 (18), 75 (32), 69 (19), 65 (11), 63 (31), 59 (32), 53 (23), 51 (48), 45 (38), 39 (14).

7-Nitro-2H-1,5-benzodioxepin-3(4H)-one (15). As described in Exper. 3, with 11d (0.81 g, 3.03 mmol) for 2.5 h. The obtained brown semi-solid (0.28 g) was purified by semi-prep. HPLC (t_R 11.7 min; λ_{max} 239 and 311 nm): 15 (16 mg, 2%). Orange oil. IR (KBr): 3452w, 3080w, 2925w, 2853w, 1801w, 1736s, 1587m, 1517s, 1491m, 1429w, 1346s, 1315m, 1274s, 1176w, 1122w, 1098w, 1080w, 1036m.

1H-NMR (300 MHz, CDCl₃): 8.03 – 7.11 (m, H – C(6), H – C(8), H – C(9)); 4.98 (s, CH₂(2)); 4.91 (s, CH₂(4)). 13 C-NMR (75 MHz, CDCl₃): 202.1 (C(3)); 153.7 (C(9a)); 147.9 (C(5a)); 143.3 (C(7)); 121.3 (C(9)); 119.8 (C(8)); 117.6 (C(6)); 76.3 (C(2)); 75.3 (C(4)). EI-MS: 209 (100, M^+), 193 (2), 167 (4), 166 (3), 163 (3), 151 (14), 135 (15), 120 (4), 107 (55), 105 (8), 90 (3), 79 (37), 77 (12), 75 (14), 63 (12), 51 (23). HR-ESI-MS: 209.0325 (C₉H₇O₅N⁺, M^+ ; calc. 209.0324).

12. Benzodioxepinone **15** by Procedure B. 3,4-Dihydro-7-nitro-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-1,5-benzodioxepine (**26d**). As described in Exper. 4, with **9d** (0.50 g, 3.22 mmol): **26d** (0.81 g, 85%). Deep orange viscous oil. IR (neat): 3428 (br.), 3081w, 2945s, 2870m, 2605w, 2243w, 1994w, 1809s, 1720w, 1586m, 1520s, 1494m, 1463w, 1424w, 1347s, 1322m, 1281s, 1265s, 1202w, 1182w, 1159w, 1123s, 1073s, 1033s. ¹H-NMR (300 MHz, CDCl₃): 7.80–7.77 (m, H–C(8), H–C(6)); 7.00–6.97 (m, H–C(9)); 4.71 ('t', H–C(2')); 4.54–4.27 (m, CH₂(2), H–C(3), CH₂(4)); 3.92–3.84 (m, H_a–C(6')); 3.57–3.50 (m, H_b–C(6')); 1.86–1.51 (m, CH₂(3'), CH₂(4'), CH₂(5')). ¹³C-NMR (75 MHz, CDCl₃): 155.5 (C(9a)); 149.5 (C(5a)); 142.6 (C(7)); 120.8 (C(9)); 118.8 (C(8)); 116.9 (C(6)); 98.9 (C(2')); 74.4 (C(3)); 73.0 (C(2)); 72.1 (C(4)); 62.8 (C(6')); 30.6 (C(3')); 25.1 (C(5')); 19.4 (C(4')). EI-MS: 295 (11, M⁺), 211 (19), 193 (19), 180 (1), 166 (12), 147 (3), 134 (1), 120 (4), 107 (2), 91 (3), 85 (100), 79 (5), 67 (10), 63 (3), 57 (11), 55 (4), 51 (3).

3,4-Dihydro-7-nitro-2H-1,5-benzodioxepin-3-ol (**27d**). As described in *Exper. 4*, with **26d** (0.76 g, 2.57 mmol). The obtained yellow resin was crystallized from Et₂O: **27d** (0.28 g, 52%). Pale yellow solid. M.p. (Et₂O) 74–78°. IR (KBr): 3453 (br.), 3105w, 3076w, 2953w, 1802w, 1721w, 1587w, 1518s, 1496m, 1466w, 1422w, 1351s, 1325m, 1283s, 1270m, 1186w, 1126w, 1103w, 1077m, 1020m. ¹H-NMR (300 MHz, CDCl₃): 7.87–7.82 (m, H–C(6), H–C(8)); 7.05 (d, J = 8.7, H–C(9)); 4.36–4.21 (m, CH₂(2), H–C(3), CH₂(4)); 2.12 (br. s, OH). ¹³C-NMR (75 MHz, CDCl₃): 155.9 (C(9a)); 150.1 (C(5a)); 143.2 (C(7)); 121.5 (C(9)); 119.4 (C(8)); 117.5 (C(6)); 74.9 (C(2)); 74.6 (C(4)); 69.0 (C(3)). EI-MS: 211 (100, M⁺), 192 (2), 180 (14), 167 (11), 166 (11), 155 (9), 152 (4), 137 (12), 134 (5), 121 (11), 107 (20), 94 (4), 91 (6), 79 (15), 77 (4), 75 (4), 65 (16), 63 (16), 57 (15), 51 (13).

7-Nitro-2H-1,5-benzodioxepin-3(4H)-one (15). At -78° under N_2 , DMSO (0.42 g, 5.37 mmol) in CH₂Cl₂ (5 ml) was added dropwise over 20 min to oxalyl chloride (0.34 g, 2.68 mmol) in CH₂Cl₂ (40 ml), and the mixture was stirred for 15 min. Then **27d** (0.25 g, 1.18 mmol) in CH₂Cl₂ (5 ml) was added dropwise over 30 min followed by stirring at -78° for 90 min. Et₃N (0.68 g, 6.72 mmol) was then added dropwise over 15 min, and the soln. was stirred for a further 2 h at -78° . THF/H₂O 1:1 (10 ml) was added while maintaining the temp. of the mixture below -60° . The soln. was then allowed to reach 0° followed by addition of CH₂Cl₂ (40 ml). The org. mixture was washed with 2m HCl (2 × 60 ml), dried (MgSO₄), and concentrated and the crude resin crystallized from MeCN: **15** (0.21 g, 84%). Bright yellow semi-solid. M.p. (MeCN) 110–112°.

13. Benzodioxepinone **16** by Procedure A. 2,2'-[(4-Formyl-1,2-phenylene)bis(oxy)]bis[acetic Acid] Dimethyl Ester (**10e**). As described in Exper. 3 (microwave procedure), with **9e** (1.00 g, 7.24 mmol). Purification by bulb-to-bulb distillation (210°/0.1 Torr) gave **10e** (1.61 g, 79%). White solid. The conventional method gave **10e** in 55% yield. M.p. (pentane) 84–86°. IR (KBr): 2957w, 1765m, 1751s, 1682m, 1598m, 1588m, 1520m, 1442m, 1428m, 1404w, 1392w, 1380w, 1340m, 1332m, 1290m, 1218s, 1175m, 1133s, 1080w, 1060m. ¹H-NMR (300 MHz, CDCl₃)¹): 9.83 (s, CHO); 7.50–6.91 (m, H–C(3'), H–C(5'), H–C(6')); 4.81 (s, CH₂(1'')); 4.78 (s, CH₂(2)); 3.80 (s, Me(3)O, Me(3'')O). ¹³C-NMR (75 MHz, CDCl₃)¹): 190.4 (CHO)); 168.4 (C(1)O, C(2'')O); 152.8 (C(1')); 148.1 (C(2')); 131.0 (C(4')); 126.9 (C(5')); 113.3 (C(6')); 112.6 (C(3')); 65.9 (CH₂(1''), CH₂(2)); 52.3 (Me(3)O, Me(3'')O). EI-MS: 282 (38, *M*⁺), 250 (11), 223 (7), 193 (4), 177 (5), 163 (18), 149 (35), 135 (4), 119 (8), 105 (5), 95 (6), 77 (11), 63 (8), 51 (10), 45 (100).

Methyl 7-Formyl-3,4-dihydro-3-oxo- and 8-Formyl-3,4-dihydro-3-oxo-2H-1,5-benzodioxepine-2-carboxylate (11e). As described in Exper. 3, with 10e (0.30 g, 1.06 mmol): crude 11e (0.13 g, 50%). Yellow oil as a complex mixture. IR (neat): 3425m, 2956m, 2361w, 1742s, 1688m, 1599m, 1509m, 1440m, 1276s,

1212s, 1163m, 1130m, 1063m. EI-MS: 178 (56), 149 (100), 121 (25), 103 (1), 91 (6), 79 (6), 63 (14), 51 (12), 42 (8).

3,4-Dihydro-3-oxo-2H-1,5-benzodioxepine-7-carboxaldehyde (16). As described in Exper. 3, with 11e (0.17 g, 0.68 mmol). Extraction was performed without NaHCO3 neutralization: yellow oily 16 (0.11 g, 85%). The sample was purified by semi-prep. HPLC ($t_{\rm R}$ 7.8 min, $\lambda_{\rm max}$ 270 nm): 16 (12 mg, 11%). Yellow semi-solid. IR (KBr): 3437 (br.), 3041w, 2923w, 2852w, 2836w, 2735w, 1802w, 1736s, 1704w, 1683s, 1605w, 1575s, 1508m, 1438w, 1421w, 1383w, 1359w, 1341w, 1314m, 1299m, 1283s, 1269s, 1236w, 1208w, 1162w, 1109w, 1044m, 1027w, 1017w. ¹H-NMR (300 MHz, CDCl3): 9.85 (s, CHO); 7.53-7.50 (m, H-C(6), H-C(8)); 7.10 (d, J = 8.8, H-C(9)); 4.79 (s, CH2(2)); 4.71 (s, CH2(4)). ¹³C-NMR (75 MHz, CDCl3): 203.0 (C(3)); 190.2 (CHO); 153.5 (C(9a)); 148.6 (C(5a)); 132.4 (C(7)); 126.0 (C(8)); 122.7 (C(9)); 121.6 (C(6)); 76.1 (C(2)); 75.2 (C(4)). EI-MS: 192 (100, M⁺), 163 (81), 149 (6), 135 (13), 121 (5), 108 (15), 91 (4), 79 (26), 63 (12), 51 (19), 42 (6), 39 (5).

14. Preparation of **16** by Procedure B. 3,4-Dihydro-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-1,5-benzodioxepine-7-carboxaldehyde (**26e**). As described in Exper. 4, with **9e** (0.50 g, 3.62 mmol): **26e** (0.74 g, 74%). Caramel brown viscous oil. IR (neat): 3422 (br.), 3063w, 2943m, 2869m, 2733w, 1803w, 1686s, 1601s, 1573m, 1501s, 1432m, 1389w, 1356w, 1316m, 1281s, 1201w, 1158m, 1127s, 1074s, 1033s.

1H-NMR (300 MHz, CDCl₃): 9.77 (s, CHO); 7.42–6.95 (m, H–C(6), H–C(8), H–C(9)); 4.72 ('t', H–C(2')); 4.59–4.22 (m, CH₂(4), H–C(3), CH₂(2)); 4.01–3.78 (m, H_a–C(6')); 3.56–3.51 (m, H_b–C(6')); 1.80–1.18 (m, CH₂(5'), CH₂(4'), CH₂(3')).

13C-NMR (75 MHz, CDCl₃): 190.6 (CHO); 155.3 (C(9a)); 150.1 (C(5a)); 131.7 (C(7)); 125.2 (C(8)); 121.4 (C(9)); 121.3 (C(6)); 98.7 (C(2')); 74.0 (C(3)); 72.9 (C(2)); 72.0 (C(4)); 62.8 (C(6')); 30.7 (C(3')); 25.2 (C(5')); 19.5 (C(4')). EI-MS: 278 (22, M+), 250 (1), 222 (2), 194 (30), 176 (19), 164 (16), 149 (30), 137 (4), 121 (3), 119 (4), 113 (3), 109 (2), 105 (1), 103 (1), 99 (1), 93 (1), 91 (5), 85 (100), 79 (5), 67 (13), 57.1 (14), 55 (7), 51 (3).

3,4-Dihydro-3-hydroxy-2H-1,5-benzodioxepine-7-carboxaldehdye (27e). As described in Exper. 4, with 26e (0.56 g, 2.01 mmol): 27e (0.35 g, 89%). Yellow resin. IR (neat): 3440 (br.), 3078w, 2939m, 2856m, 2735w, 2620w, 1794w, 1731m, 1688s, 1599s, 1588m, 1574m, 1505s, 1434s, 1394w, 1359w, 1338w, 1317m, 1280s, 1169m, 1138m, 1099w, 1035m. 1 H-NMR (300 MHz, CDCl₃): 9.82 (s, CHO); 7.47 – 7.45 (m, H–C(6), H–C(8)); 7.06 (d, J=8.8, H–C(9)); 4.32 – 4.14 (m, CH₂(2), H–C(3), CH₂(4)); 2.87 (br. s, OH). 1 3C-NMR (75 MHz, CDCl₃): 190.6 (CHO); 155.8 (C(9a)); 150.8 (C(5a)); 132.2 (C(7)); 125.7 (C(8)); 122.6 (C(9)); 121.8 (C(6)); 74.7 (C(2)); 74.5 (C(4)); 69.1 (C(3)). EI-MS: 194 (100, M⁺), 175 (3), 163 (17), 149 (65), 137 (21), 121 (13), 119 (6), 109 (6), 105 (3), 95 (6), 91 (5), 79 (8), 77 (8), 65 (9), 63 (11), 57 (7), 51 (9).

3,4-Dihydro-3-oxo-2H-1,5-benzodioxepine-7-carboxaldehyde (16). As described in Exper. 4, with 27e (0.25 g, 1.29 mmol). The crude resin was crystallized from MeCN: 16 (0.13 g, 52%). White solid. M.p. (MeCN) $110-112^{\circ}$.

15. Benzodioxepinone 17 by Procedure A. 2,2'-{[4-(tert-Butyl)-1,2-phenylene]bis(oxy)]bis[acetic Acid] Dimethyl Ester (10f). As described in Exper. 3 (conventional method), with 9f (1.00 g, 6.02 mmol). Bulb-to-bulb distillation (190°/0.3 Torr) performed twice afforded 10f (1.40 g, 75%). Clear oil. IR (neat): 2908s, 2956s, 2870m, 2111w, 1762s, 1608m, 1592m, 1583m, 1510s, 1440s, 1414m, 1376m, 1364m, 1294s, 1201s, 1153s, 1110m, 1068s, 1028m, 1005m. ¹H-NMR (300 MHz, CDCl₃)¹): 6.96 – 6.77 (m, H – C(3'), H – C(5'), H – C(6')); 4.72 (s, CH₂(1")); 4.68 (s, CH₂(2)); 3.78 (s, Me(3)O, Me(3")O); 1.26 (s, 'Bu). ¹³C-NMR (75 MHz, CDCl₃)¹): 169.5 (CO(1), CO(2")); 147.3 (C(1')); 145.9 (C(2'), C(4')); 119.4 (C(5')); 114.9 (C(6')); 114.2 (C(3')); 67.2 (CH₂(2)); 66.7 (CH₂(1")); 52.1 (Me(3)O); 52.0 (Me(3")O); 34.3 (Me₃C); 31.3 (Me₃C). EI-MS: 310 (23, M⁺), 295 (100), 222 (2), 205 (1), 191 (1), 177 (4), 163 (11), 149 (3), 133 (5), 121 (2), 117 (2), 105 (4), 91 (6), 77 (5), 65 (2), 59 (2), 51 (1), 45 (32).

Methyl 7-(tert-*Butyl*)-3,4-dihydro-3-oxo- and 8-(tert-*Butyl*)-3,4-dihydro-3-oxo-2H-1,5-benzodioxe-pine-2-carboxylate (11f). As described in *Exper.* 3, with 10f (2.42 g, 7.81 mmol): 11f (1.89 g, 87%). Clear brown oil. IR (neat): 3456m, 2958s, 2870s, 2566m, 2085s, 1741s, 1582m, 1504s, 1473m, 1437s, 1414s, 1364m, 1269s, 1206s, 1149s, 1126s, 1102m, 1079m. H-NMR (300 MHz, CDCl₃): *Regioisomer 1*: 7.11−6.94 (m, H−C(6), H−C(8), H−C(9)); 5.40 (s, H−C(2)); 4.82 (s, s = 16.4, H_a−C(4)); 4.60 (s, s = 16.6, H−C(2)); 3.87 (s, MeO); 1.28 (s, s = 10); *Regioisomer 2*: 7.11−6.94 (s, H−C(6), H−C(8), H−C(9)); 5.36 (s, H−C(2)); 4.81 (s, s = 17.1, H_a−C(4)); 4.58 (s, s = 16.8, H_b−C(4)); 3.86 (s, MeO); 1.28 (s, s = 10.13 C-NMR (75 MHz, CDCl₃): *Regioisomer 1*: 199.6 (C(3)); 165.5 (CO₂); 148.5 (C(9a)); 146.4 (C(5a));

144.5 (C(7)); 120.8 (C(8)); 119.8 (C(9)); 117.5 (C(6)); 81.7 (C(2)), 76.7 (C(4)); 52.9 (MeO); 34.4 (Me₃C); 31.3 (Me_3 C); $Regioisomer\ 2$: 199.6 (C(3)); 165.4 (CO₂); 147.2 (C(9a)); 145.7 (C(5a)); 144.8 (C(7)); 121.8 (C(8)); 120.6 (C(9)); 118.2 (C(6)); 81.9 (C(4)); 76.8 (C(2)); 52.9 (MeO); 34.2 (Me₃C); 31.3 (Me_3 C). EI-MS: 278 (22, M^+), 263 (100), 235 (5), 219 (3), 205 (6), 191 (3), 175 (5), 163 (3), 147 (4), 133 (2), 121 (3), 115 (4), 105 (3), 91 (6), 77 (5), 65 (2), 55 (2), 41 (2), 32 (5).

7-(tert-*Butyl*)-2H-*1*,5-benzodioxepin-3(4H)-one (17). As described in *Exper.* 3, with 11f (1.24 g, 4.46 mmol) for 2.5 h, with the aq. workup involving neutralization with aq. NaHCO₃ soln. prior to extraction. The obtained brown oil was bulb-to-bulb distilled at 100–105°/0.2 Torr: 17 (0.82 g, 84%). White crystalline solid. M.p. (pentane) 66–68°. IR (KBr): 2965s, 2950*m*, 2901*m*, 2868*m*, 1791*w*, 1742*s*, 1577*m*, 1504*s*, 1477*m*, 1461*m*, 1438*m*, 1424*m*, 1409*m*, 1394*w*, 1367*w*, 1359*w*, 1347*w*, 1300*s*, 1274*s*, 1248*m*, 1206*m*, 1189*m*, 1127*m*, 1098*w*, 1045*s*. ¹H-NMR (300 MHz, CDCl₃): 7.02–6.79 (*m*, H–C(6), H–C(8), H–C(9)); 4.72 (*s*, CH₂(2)); 4.69 (*s*, CH₂(4)); 1.26 (*s*, 'Bu). ¹³C-NMR (75 MHz, CDCl₃): 204.9 (C(3)); 147.6 (C(9a)); 147.4 (C(5a)); 145.8 (C(7)); 120.7 (C(8)); 120.3 (C(9)); 117.8 (C(6)); 75.7 (C(2)); 75.6 (C(4)); 34.3 (Me₃C); 31.3 (*Me*₃C). EI-MS: 220 (26, *M*⁺), 205 (100), 177 (11), 149 (3), 135 (4), 121 (4), 105 (4), 91 (7), 77 (7), 6.5 (3), 55 (2), 41 (3).

16. Benzodioxepinone 17 by Procedure B. 7-(tert-Butyl)-3,4-dihydro-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-I,5-benzodioxepine (26f). As described in Exper. 4, with 9f (0.5 g, 3.01 mmol): 26f (0.90 g, 98%). Light brown viscous oil. IR (neat): 3435 (br.), 3039w, 2952s, 2870m, 2741w, 2662w, 1727w, 1625w, 1613w, 1574m, 1505s, 1463m, 1455m, 1442w, 1412w, 1393w, 1363w, 1355w, 1306s, 1274m, 1263m, 1239m, 1202m, 1185w, 1161w, 1122s, 1097w, 1076s, 1034s, 1023m. 1 H-NMR (300 MHz, CDCl₃): 6.96–6.86 (m, H–C(6), H–C(8), H–C(9)); 4.74 (t, J=2.8, H–C(2')); 4.46–4.21 (m, CH₂(2), H–C(3), CH₂(4')); 3.95–3.86 (m, H_a–C(6')); 3.57–3.49 (m, H_b–C(6')); 1.90–1.43 (m, CH₂(3'), CH₂(4'), CH₂(5')); 1.26 (m, CH₂(5)); 74.5 (C(3)); 73.0 (C(2)); 71.9 (C(4)); 62.6 (C(6')); 34.1 (Me₃C); 31.3 (Me₃C); 30.8 (C(3')); 25.3 (C(5')); 19.4 (C(4')). EI-MS: 306 (77, m+), 291 (58), 263 (3), 261 (2), 250 (2), 248 (1), 235 (13), 222 (19), 207 (81), 204 (5), 193 (5), 189 (24), 177 (12), 163 (6), 161 (6), 151 (10), 147 (3), 133 (10), 121 (4), 119 (4), 117 (3), 115 (3), 105 (12), 91 (12), 85 (100), 77 (11), 67 (17), 57 (25), 55 (17), 51 (36).

7-(tert-*Butyl*)-3,4-dihydro-2H-1,5-benzodioxepin-3-ol (**27f**). As described in *Exper.* 4, with **26f** (0.80 g, 2.61 mmol): **27f** (0.47 g, 81%). White solid. M.p. (pentane) $48-50^{\circ}$. IR (KBr): 3459 (br.), 2962m, 2866w, 1578w, 1503s, 1460w, 1447w, 1409m, 1390w, 1364w, 1311s, 1274s, 1252m, 1236s, 1204w, 1184w, 1118m, 1090m, 1057m, 1042s, 1022m. ¹H-NMR (300 MHz, CDCl₃): 7.04–6.95 (m, H–C(6), H–C(8), H–C(9)); 4.32–4.23 (m, H_a–C(2), H_a–C(4)); 4.09–4.04 (m, H_b–C(2), H–C(3), H_b–C(4)); 2.70 ('s', OH); 1.27 (s, 'Bu). ¹³C-NMR (75 MHz, CDCl₃): 150.4 (C(9a)); 148.5 (C(5a)); 147.5 (C(7)); 120.9 (C(8)); 120.8 (C(9)); 118.6 (C(6)); 74.7 (C(2), C(4)); 69.7 (C(3)); 34.2 (Me₃C); 31.3 (Me_3 C). EI-MS: 222 (29, M^+), 207 (100), 163 (4), 151 (9), 133 (5), 123 (5), 105 (7), 91 (5), 77 (6), 65 (2).

7-(tert-*Butyl*)-2H-1,5-benzodioxepin-3(4H)-one (17). As described in *Exper.* 4, with 27f (0.41 g, 1.85 mmol). The obtained clear oil was purified by bulb-to-bulb distillation at $100-105^{\circ}/0.2$ Torr: 17 (0.30 g, 73%). Crystalline solid M.p. (pentane) $66-68^{\circ}$.

17. Benzodioxepinone **18** by Procedure A. 2,2'-[Naphthalene-2,3-diylbis(oxy)]bis[acetic Acid] Dimethyl Ester (**10g**). As described in Exper. 3 (conventional method), with **9g** (2.00 g, 12.49 mmol). Purification was achieved by continuous trituration with CH₂Cl₂ until an ivory solid appeared: **10g** (3.56 g, 96%). The microwave procedure gave **10g** in 75% yield. M.p. (CH₂Cl₂) 78–80°. IR (KBr): 2975w, 2953m, 1751s, 1737s, 1629w, 1603w, 1585w, 1510m, 1488s, 1460m, 1447m, 1436m, 1410w, 1380m, 1339w, 1284s, 1247s, 1228s, 1189m, 1168s, 1123m, 1078w, 1047m, 1034w, 1024m. ¹H-NMR (300 MHz, CDCl₃)¹): 7.67 (dd, *J* = 6.1, 3.1, H-C(5'), H-C(8')); 7.35 (dd, *J* = 6.1, 3.5, H-C(6'), H-C(7')); 7.12 (s, H-C(1'), H-C(4')); 4.83 (s, CH₂(2), CH₂(1")); 3.82 (s, Me(3)O, Me(3")O). ¹³C-NMR (75 MHz, CDCl₃)¹): 169.1 (C(1)O, C(2")O); 147.7 (C(2'), C(3')); 129.5 (C(4a), C(8a)); 126.5 (C(5'), C(8')); 124.8 (C(6'), C(7')); 109.6 (C(1'), C(4')); 66.2 (CH₂(2), CH₂(1")); 52.3 (Me(3)O, MeO(3")). EI-MS: 304 (100, *M*⁺), 272 (8), 243 (23), 213 (8), 203 (5), 185 (26), 171 (87), 147 (6), 127 (26), 115 (36), 102 (25), 88 (6), 76 (6), 63 (6), 45 (67).

3,4-Dihydro-methyl-3-oxo-2H-naphtho[2,3-b][1,4]dioxepine-2-carboxylate (11g). As described in Exper. 3, with 10g (1.39 g, 4.57 mmol): 11g (1.16 g, 93%). Caramel brown oil. A portion of this sample was purified by semi-prep. HPLC (t_R 29.9 min, λ_{max} 230 nm): 11g (21 mg, 2%). Light brown oil. IR (KBr):

3453m, 3056w, 2954w, 1744s, 1635w, 1602w, 1505s, 1469s, 1363m, 1277s, 1225s, 1172s, 1117m, 1072m, 1024m. ¹H-NMR (300 MHz, CDCl₃): 7.70 (dd, J = 6.0, 3.2, H-C(7), H-C(10)); 7.54 (s, H-C(11)); 7.45 (s, H-C(6)); 7.39 (dd, J = 6.7, 3.1, H-C(8), H-C(9)); 5.41 (s, H-C(2)); 4.96 (d, J = 17.1, H_a-C(4)); 4.81 (d, J = 17.1, H_b-C(4)); 3.86 (s, MeO). ¹³C-NMR (75 MHz, CDCl₃): 199.5 (C(3)); 165.3 (CO₂); 147.2 (C(11a)); 146.5 (C(5a)); 131.0 (C(10a)); 130.0 (C(6a)); 125.7 (C(8)); 125.3 (C(9)); 126.7 (C(10)); 126.8 (C(7)); 118.0 (C(11)); 116.7 (C(6)); 81.7 (C(2)); 76.5 (C(4)); 53.0 (MeO). EI-MS: 272 (s, s), 243 (12), 213 (s), 185 (17), 171 (61), 157 (2), 147 (3), 127 (19), 115 (28), 102 (20), 88 (s), 76 (s), 63 (s), 45 (100).

2H-Naphtho[2,3-b][1,4]dioxepin-3(4H)-one (18). As described in *Exper. 3*, with 11g (0.33 g, 1.21 mmol) in a minimal amount of EtOH/AcOEt 2:1 for 2.5 h. Quenching of the reaction soln. in aq. NaHCO₃ soln. and usual workup gave brown semi-solid 18 (0.17 g, 65%). A portion of the sample was purified by semi-prep. HPLC (t_R 15.1 min, λ_{max} 230 nm): 11g (18 mg, 7%). Light brown crystalline solid. M.p. (MeCN) 70–72°. IR (KBr): 3442m, 2960s, 2925s, 2854s, 1737m, 1599w, 1504m, 1470m, 1413w, 1363m, 1330w, 1289m, 1262s, 1169w, 1151w, 1098m, 1046s, 1030s. ¹H-NMR (300 MHz, CDCl₃): 7.51 (dd, J = 6.4, 3.2, H – C(7), H – C(10)); 7.19 (dd, J = 6.4, 3.1, H – C(8), H – C(9)); 7.07 (s, H – C(6), H – C(11)); 4.60 (s, CH₂(2), CH₂(4)). ¹³C-NMR (75 MHz, CDCl₃): 204.8 (C(3)); 148.2 (C(5a), C(11a)); 130.4 (C(6a), C(10a)); 126.7 (C(7), C(10)); 125.4 (C(8), C(9)); 117.2 (C(11), C(6)); 75.4 (C(2), C(4)). EI-MS: 214 (100, M⁺), 186 (6), 171 (58), 160 (4), 142 (2), 130 (33), 114 (32), 102 (63), 88 (12), 76 (15), 63 (20), 51 (16), 39 (7). HR-ESI-MS: 215.0715 (C₁₃H₁₀O⁺₃, [M + 1]⁺; calc. 215.0629).

18. Benzodioxepinone **18** by Procedure B. 3,4-Dihydro-3-[(tetrahydro-2H-pyran-2-yl)oxy]-2H-naphtho[2,3-b][1,4]dioxepine (**26g**). As described in Exper. 4, with **9g** (0.5 g, 3.12 mmol): **26g** (0.77 g, 82%). Brown viscous oil. IR (neat): 3331 (br.), 3056w, 2943s, 2870m, 1708m, 1670s, 1631w, 1602w, 1503s, 1472s, 1453s, 1386m, 1359m, 1290s, 1260s, 1247s, 1201m, 1172m, 1159m, 1125s, 1074s, 1031s. ¹H-NMR (300 MHz, CDCl₃): 7.67-7.61 (m, H-C(7), H-C(10)); 7.36 (s, H-C(6), H-C(11)); 7.33-7.30 (m, H-C(8), H-C(9)); 4.77 ('t', H-C(2')); 4.51-4.30 (m, CH₂(2), H-C(3), CH₂(4)); 4.05-3.83 (m, H_a-C(6')); 3.59-3.44 (m, H_b-C(6')); 1.90-1.20 (m, CH₂(3'), CH₂(4'), CH₂(5')). ¹³C-NMR (75 MHz, CDCl₃): 151.1 (C(11a)); 150.2 (C(5a)); 130.2 (C(6a), C(10a)); 126.5 (C(7), C(10)); 124.8 (C(8), C(9)); 116.9 (C(6), C(11)); 98.5 (C(2')); 74.6 (C(3)); 73.0 (C(2)); 71.9 (C(4)); 62.7 (C(6')); 30.7 (C(3')); 25.3 (C(5')); 19.4 (C(4')). EI-MS: 300 (65, M⁺), 272 (14), 244 (8), 216 (31), 198 (17), 185 (10), 181 (2), 171 (44), 169 (14), 160 (18), 141 (9), 131 (11), 128 (11), 127 (14), 115 (22), 114 (17), 102 (22), 85 (100), 77 (4), 76 (4), 67 (18), 57 (28), 55 (11).

3,4-Dihydro-2H-naphtho[2,3-b][1,4]dioxepin-3-ol (27g). As described in Exper. 4, with 26g (0.65 g, 2.16 mmol). Crystallization of the residue from 'BuOMe yielded 27g (0.34 g, 73%). Red solid. M.p. ('BuOMe) $79-80^\circ$. IR (KBr): 3420 (br.), 3056w, 2925w, 1777w, 1726m, 1627w, 1598m, 1503s, 1474m, 1458w, 1445w, 1432w, 1394w, 1360m, 1341w, 1289m, 1263s, 1173m, 1151w, 1109m, 1069w, 1031s. ¹H-NMR (300 MHz, CDCl₃): 7.71–7.68 (m, H–C(7), H–C(10)); 7.46 (s, H–C(6), H–C(11)); 7.39–7.36 (m, H–C(8), H–C(9)); 4.41–4.36 (m, H_a–C(2), H_a–C(4)); 4.14–4.09 (m, H_b–C(2), H–C(3), H_b–C(4)); 2.30 ('s', OH). ¹³C-NMR (75 MHz, CDCl₃): 151.1 (C(5a), C(11a)); 130.7 (C(6a), C(10a)); 126.8 (C(7), C(10)); 125.3 (C(8), C(9)); 118.1 (C(6), C(11)); 75.3 (C(2), C(4)); 69.8 (C(3)). EI-MS: 216 (100, M^+), 197 (1), 185 (13), 172 (16), 171 (32), 160 (33), 145 (5), 131 (16), 127 (13), 115 (22), 114 (23), 102 (13), 88 (5), 77 (4), 76 (4), 63 (4).

2H-Naphtho[2,3-b][1,4]dioxepin-3(4H)-one (18). As described in Exper. 4, with 27g (0.31 g, 1.43 mmol): 18 (0.19 g, 61%). Grey-white solid. M.p. (MeCN) 70-74°.

19. 7-Bromo-8-methyl-2H-1,5-benzodioxepin-3(4H)-one (19). Sodium bromate (2.54 g, 16.83 mmol) was added to 1 (1.00 g, 5.62 mmol) dissolved in cyclohexane/ H_2O 4:3 (70 ml). Then sodium metabisulfite (0.98 g, 5.15 mmol) in H_2O (10 ml) was added dropwise, and the mixture was stirred at 50° for 8 h. The deep yellow soln. was quenched with ice/ H_2O (250 ml) and extracted with E_2O (3 × 100 ml) and the combined org. phase washed with H_2O (2 × 200 ml), dried (MgSO₄), and concentrated: 19 (1.23 g, 85%). Pale yellow powdery solid. M.p. (pentane) 88 – 90°. IR (KBr): 2930w, 2900w, 1737s, 1602w, 1566m, 1480s, 1451m, 1428m, 1416m, 1378w, 1294s, 1266m, 1230w, 1178m, 1152s, 1052s, 1036s. 1 H-NMR (300 MHz, CDCl₃): 7.17 (s, H–C(6)); 6.87 (s, H–C(9)); 4.67 (s, CH₂(2)); 4.66 (s, CH₂(4)); 2.28 (s, Me). 13 C-NMR (75 MHz, CDCl₃): 203.8 (C(3)); 147.3 (C(9a)); 146.6 (C(5a)); 133.3 (C(8)); 124.2 (C(6)); 122.3 (C(9)); 117.4 (C(7)); 75.7 (C(2)); 75.5 (C(4)); 21.9 (Me). EI-MS: 258 (89, M^+), 257 (10), 256 (92), 215 (12), 213

(12), 202 (3), 200 (4), 174 (25), 172 (28), 149 (30), 135 (5), 121 (6), 105 (3), 93 (100), 77 (18), 65 (14), 51 (12), 39 (10). HR-ESI-MS: 256.9812 ($C_{10}H_{9}O_{3}Br^{+}$, M^{+} ; calc. 256.9813).

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